



# हेमचंद यादव विश्वविद्यालय, दुर्ग (छ.ग.)

(पूर्व नाम- दुर्ग विश्वविद्यालय, दुर्ग)

रायपुर नाका दुर्ग (छ.ग.)-491001

ई मेल : [academic@durguniversity.ac.in](mailto:academic@durguniversity.ac.in)

वेब साइट : [www.durguniversity.ac.in](http://www.durguniversity.ac.in)

दूरभाष : 0788-2359400

क्र. 2960/A / अका. / 2020

दुर्ग, दिनांक 10/9/2020

प्रति,

प्राचार्य,  
समस्त संबद्ध महाविद्यालय,  
हेमचंद यादव विश्वविद्यालय,  
दुर्ग (छ.ग.)

विषय:- स्नातक स्तर भाग-दो के पाठ्यक्रम विषयक।

संदर्भ:- संयुक्त संचालक, उच्च शिक्षा विभाग के पत्र क्र. 2456/315/आउशि/सम/2019, दिनांक 16.05.2019।

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विषयांतर्गत लेख है कि संदर्भित पत्र के माध्यम से प्राप्त स्नातक स्तर भाग-दो के निम्नलिखित कक्षा/विषयों के परिवर्तित/संशोधित पाठ्यक्रम शिक्षा सत्र 2020-21 से लागू किये जाते हैं:-

1. बी.ए. - आधार पाठ्यक्रम-हिन्दी भाषा, हिन्दी साहित्य, राजनीतिशास्त्र, अर्थशास्त्र, नृत्य, दर्शनशास्त्र, समाजशास्त्र, इतिहास, मानवविज्ञान, संस्कृत, सांख्यिकी प्राचीन भारतीय इतिहास, भूगोल, मनोविज्ञान
2. बी.एस-सी.- आधार पाठ्यक्रम-हिन्दी भाषा, जीव विज्ञान, मानवविज्ञान, बायोटेक्नोलॉजी, कम्प्यूटर साईंस, गणित, भौतिक शास्त्र, प्राणीशास्त्र, सूक्ष्मजीव विज्ञान, वनस्पतिशास्त्र, भूविज्ञान, इलेक्ट्रॉनिक्स, रसायन शास्त्र, सांख्यिकी, भूगोल।
3. बी.ए./बी.एस.सी (गृह विज्ञान) - आधार पाठ्यक्रम - हिन्दी भाषा एवं गृह विज्ञान।

उपरोक्त विषयों को शिक्षा सत्र 2020-21 से संशोधित रूप में स्नातक स्तर भाग-दो के लिए लागू किया जाता है स्नातक स्तर भाग-एक हेतु सत्र 2019-20 में लागू पाठ्यक्रम मान्य होंगे एवं भाग - तीन के पाठ्यक्रम यथावत रहेंगे।

टीप:- परिवर्तित/संशोधित पाठ्यक्रम विश्वविद्यालय के परीक्षा विभाग एवं वेबसाइट पर प्रकाशित करने हेतु वेबसाइट प्रभारी को उपलब्ध करा दी गई है।

कुलसचिव

क्र. 2961/A / अका. / 2020

दुर्ग, दिनांक 10/9/2020

प्रतिलिपि:-

1. संयुक्त संचालक, उच्च शिक्षा विभाग के पत्र क्र. 2456/315/आउशि/सम/2019, दिनांक 16.05.2019 परिपेक्ष्य में सूचनार्थ।
2. उपकुलसचिव, परीक्षा विभाग एवं उपकुलसचिव, गोपनीय विभाग हेमचंद यादव विश्वविद्यालय, दुर्ग।
3. कुलपति के निज सहायक एवं कुलसचिव के निज सहायक, हेमचंद यादव विश्वविद्यालय, दुर्ग।

सहा. कुलसचिव (अका.)

**REVISED ORDINANCE NO. 21**  
**BACHELOR OF SCIENCE**

1. The three year course has been broken up into three Parts. Part-I known as B.Sc. Part-I examination at the end of the first year, Part-II known as B.Sc. Part-II examination at the end of the second year and Part-III known as B.Sc. Part-III examination at the end of the third year.
2. A candidate who after passing (10+2) Higher Secondary or Intermediate examination of C.G. Board of Secondary Education Bhopal or any other Examination recognised by the University or C.G. Board of Secondary Education as equivalent thereto, has attended a regular course of study in an affiliated College or in the Teaching Department of the University for one academic year shall be eligible for appearing at the B.Sc. Part-I examination.
3. A candidate who, after passing the B.Sc.-I examination of the University or any other examination recognised by the University as equivalent thereto, has attended a regular course of study for one academic year in an affiliated college or in the Teaching Department of the University shall be eligible for appearing at the B.Sc. Part-II examination.
4. A candidate who, after passing the B.Sc. Part-II examination of the University, has completed a regular course of study for one academic year in an affiliated college or in the Teaching Department of the University shall be eligible for appearing at the B.Sc. Part-III examination.
5. Besides regular students, subject to their compliance with this Ordinance ex-student and non-collegiate candidates shall be permitted to offer only such subjects/papers as are taught to the regular student at any of the University Teaching Department or College.
6. Every candidate appearing in B.Sc. Part-I, Part-II and Part-III examination shall be examined in-
  - (i) Foundation Course:
  - (ii) Any one of the following combinations of three subjects:-
    1. Physics, Chemistry & Mathematics.
    2. Chemistry, Botany & Zoology.
    3. Chemistry, Physics & Geology.
    4. Chemistry, Botany & Geology.
    5. Chemistry, Zoology & Geology.
    6. Geology, Physics & Mathematics.
    7. Chemistry, Mathematics & Geology.
    8. Chemistry, Botany & Defence Studies.
    9. Chemistry, Zoology & Defence Studies.
    10. Physics, Mathematics & Defence Studies.
    11. Chemistry, Geology & Defence Studies.

12. Physics, Mathematics & Statistics
  13. Physics, Chemistry & Statistics
  14. Chemistry, Mathematics & Statistics.
  15. Chemistry, Zoology & Anthropology.
  16. Chemistry, Botany & Anthropology.
  17. Chemistry, Geology & Anthropology.
  18. Chemistry, Mathematics & Statistics.
  19. Chemistry, Anthropology & Defence Studies.
  20. Geology, Mathematics & Statistics.
  21. Mathematics, Defence Studies & Statistics
  22. Anthropology, Mathematics & Statistics
  23. Chemistry, Anthropology & Applied Statistics
  24. Zoology, Botany & Anthropology
  25. Physics, Mathematics & Electronics.
  26. Physics, Mathematics & Computer Application
  27. Chemistry, Mathematics & Computer Application
  28. Chemistry, Bio-Chemistry & Pharmacy
  29. Chemistry, Zoology & Fisheries.
  30. Chemistry, Zoology & Agriculture
  31. Chemistry, Zoology & Sericulture
  32. Chemistry, Botany & Environmental Biology
  33. Chemistry, Botany & Microbiology
  34. Chemistry, Zoology & Microbiology
  35. Chemistry, Industrial Chemistry & Mathematics
  36. Chemistry, Industrial Chemistry & Zoology
  37. Chemistry, Biochemistry, Botany
  38. Chemistry, Biochemistry, Zoology
  39. Chemistry, Biochemistry, Microbiology
  40. Chemistry, Biotechnology, Botany
  41. Chemistry, Biotechnology, Zoology
  42. Geology, Chemistry & Geography
  43. Geology, Mathematics & Geography
  44. Mathematics, Physics & Geography
  45. Chemistry, Botany & Geography
- (iii) Practical in case prescribed for coresubjects.

7. Any candidate who has passed the B.Sc. examination of the University shall be allowed to present himself for examination in any of the additional subjects prescribed for the B.Sc. examination and not taken by him at the degree examination. Such candidate will have to first appear and pass the B.Sc. Part-I examination in the subjects which he proposes to offer and then the B.Sc. Part-II and Part-III examination in the same subject. Successful candidates will be given a certificate to that effect.

8. In order to pass at any part of the three year degree course examination an examinee must obtain not less than 33% of the total marks in each subject/ group of subjects. In subject/ group of subjects where both theory and practical examination are provided an examinee must pass in both theory and practical parts of the examination separately.
9. Candidate will have to pass separately at the Part-I, Part-II and Part-III examinations. No division shall be assigned on the result of the Part-I and Part-II examination. In determining the division of the final examination, total marks obtained by the examinees in their Part-I, Part-II and Part-III examination in the aggregate shall be taken in to account. Provided in case of candidate who has passed the examination through supplementary examination having failed in one subject/ group only, the total aggregate marks being carried over for determining the division shall include actual marks obtained in the subject/ group in which he appeared at the supplementary examination.
10. Successful examinee at the Part-III examination obtaining 60% or more marks shall be places in the First Division, those obtaining less than 60% but not less than 45% marks in the Second Division and other successful examinees in the Third Division.

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## SCHEME OF EXAMINATION

Subject	Paper	Max. Marks	Total Marks	Min. Marks
C Environmental Studies		75	100	33
Fild Work		25		
<b>Foundation Course</b>				
Hindi Language		75	75	26
English Language		75	75	26

**नोट:-** प्रत्येक में से 02 (दो) प्रश्न करने होंगे । सभी प्रश्न समान अंक के होंगे ।

Three Elective Subject :

1.	Physics	I	50	100	33
		II	50		
2.	Chemistry	Practical		50	17
		I	33		
		II	33	100	33
		III	34		
3.	Mathematics	Practical		50	17
		I	50		
		II	50	150	50
		III	50		
4.	Botany	I	50	100	33
		II	50		
5.	Zoology	Practical		50	17
		I	50	100	33
		II	50		
6.	Geology	Practical		50	17
		I	50	100	33
		II	50		
7.	Statistics	Practical	50		17
		I	50	100	33
		II	50		
8.	Anthropology	Practical		50	17
		I	50	100	50
		II	50		
		Practical		50	17

Subject	Paper	Max. Marks	Total Marks	Min. Marks
Compulsory Subject–Foundation Course:				
9. Defense Studies	I	50	100	33
	II	50		
	Practical			
10. MicroBiology	I	50	100	33
	II	50		
	Practical			
11. Computer Sciences	I	50	100	33
	II	50		
	Practical			
12. Information Technology	I	50	100	33
	II	50		
	Practical			
13. Industrial Chemistry	I	34	100	33
	II	33		
	III	33		
	Practical			
14. BioChemistry	I	50	100	33
	II	50		
15. BioTechnology	Practical	50	50	17
	I			
	II	50		
	Practical		50	17

### USE OF CALCULATORS

The Students of Degree/P.G. Classes will be permitted to use of Calculators in the examination hall from annual 1986 examination on the following conditions as per decision of the standing committee of the Academic Council at its meeting held on 31-1-1986.

1. Student will bring their own Calculators.
2. Calculators will not be provided either by the University or examination centres.
3. Calculators with, memory and following variables be permitted +, -, x,  $\frac{1}{x}$ , square, reciprocal, exponentials log, square root, trigonometric functions, sine, cosine, tangent etc. factorial summation, xy, yx and in the light of objective approval of merits and demerits of the viva only will be allowed.

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संशोधित पाठ्यक्रम  
बी.ए./बी.एस-सी./बी.कॉम./बी.एच.एस.-सी.  
भाग - दो, आधार पाठ्यक्रम  
प्रश्न पत्र - प्रथम (हिन्दी भाषा) (पेपर कोड - 0171)

पूर्णांक- 75

खण्ड - क निम्नलिखित 5 लेखकों के पाठ शामिल होंगे - अंक-35

1. महात्मा गांधी - चोरी और प्रायश्चित
2. आचार्य नरेंद्र देव - युवकों का समाज में स्थान
3. वासुदेव शरण अग्रवाल - मातृभूमि
4. हरि ठाकुर - डॉ. खूबचंद बघेल
5. पं. माधवराव सप्रे - सम्भाषण-कुशलता

खण्ड-ख हिन्दी भाषा और उसके विविध रूप अंक-16

1. कार्यालयीन भाषा
2. मीडिया की भाषा
3. वित्त एवं वाणिज्य की भाषा
4. मशीनी भाषा

खण्ड-ग हिन्दी की व्याकरणिक कोटियाँ अंक-24

संज्ञा, सर्वनाम, विशेषण, क्रिया विशेषण,  
समास, संधि एवं संक्षिप्तियाँ  
अनुवाद व्यवहार : अंग्रेजी से हिन्दी में अनुवाद

इकाई विभाजन-

- इकाई- 1 चोरी और प्रायश्चित : महात्मा गांधी / कार्यालयीन भाषा, मीडिया की भाषा
- इकाई- 2 युवकों का समाज में स्थान : आचार्य नरेंद्र देव / वित्त एवं वाणिज्य की भाषा, मशीनी भाषा
- इकाई- 3 मातृभूमि: वासुदेवशरण अग्रवाल / संज्ञा सर्वनाम, विशेषण, क्रिया विशेषण
- इकाई- 4 डॉ. खूबचंद बघेल : हरि ठाकुर/समास, संधि,
- इकाई- 5 सम्भाषण-कुशलता : पं. माधवराव सप्रे, / अनुवाद - अंग्रेजी से हिन्दी में अनुवाद, संक्षिप्तियाँ

मूल्यांकन योजना -

प्रत्येक इकाई से एक-एक प्रश्न पूछे जाएंगे। प्रत्येक प्रश्न में आंतरिक विकल्प होगा। प्रत्येक प्रश्न के 15 अंक होंगे। प्रत्येक इकाई को दो-दो खण्डों (क्रमशः 'क' और 'ख' में) विभक्त करते हुए निर्धारित पाठ से 8 एवं शेष पाठ्य सामग्री से 7 अंक के प्रश्न होंगे। इस प्रकार पूरे प्रश्न-पत्र के पूर्णांक 75 होंगे।

**पाठ्यक्रम संशोधन का औचित्य :** विद्यार्थी चर्चित एवं सुप्रसिद्ध व्यक्तियों के लेख के माध्यम से समाज एवं राष्ट्रहित के साथ-साथ व्यक्तित्व विकास विषयक मुद्दों से परिचित हो सकें तथा व्याकरणक एवं भाषा विषयक प्रस्तावित पाठ्यक्रम के माध्यम से हिन्दी भाषा संबंधित प्रयोग पक्ष से परिचित होते हुए प्रतियोगी परीक्षाओं की दृष्टि से ज्ञानार्जन कर सकें।

ENGLISH LANGUAGE (Paper Code-1132)

B.A. / B.Sc. /B.COM. /B.H. Sc. - II

M.M.75

The question paper for B.A. /B.Sc./B.Com./B.H.Sc., English Language and cultural values shall comprise the following units:

UNIT-I Short answer questions to be passed by (Five short answer questions of three marks each)	15 Marks
UNIT-II (a) Reading comprehension of an unseen passage (b) Vocabulary	05 Marks
UNIT-III Report-Writing	10 Marks
UNIT-IV Expansion of an idea	10 Marks
UNIT-V Grammar and Vocabulary based on the prescribed text book.	20+15Marks

Note: Question on all the units shall asked from the prescribed text which will Comprise Specimens of popular creative/writing and the following it any

- a Matter & technology
  - i. State of matter and its structure
  - ii. Technology (Electronics Communication, Space Science)
- b Our Scientists & Institutions
  - I. Life & work of our eminent scientist Arya Bhatt. Kaurd Charak Shusruta, Nagarjuna, J.C. Bose and C.V. Raman, S. Rmanujam, Homi J. Babha Birbal Sahani.
  - II. Indian Scientific Institutions (Ancient & Modern)

Books Prescribed:

Foundation English for U.G. Second Year - Published by M.P. Hindi Granth Academy, Bhopal.



**HEMCHAND YADAV VISHWAVIDYALAYA, DURG (C.G.)**  
**NEW CURRICULUM OF B.Sc. PART II**  
**SESSION 2019-20**  
**CHEMISTRY**

The new curriculum will comprise of three papers of 33, 33 and 34 marks each and practical work of 50 marks. The Curriculum is to be completed in 180 working days as per UGC norms and conforming to the directives of Govt. of Chhattisgarh. The theory papers are of 60 hrs. each duration and practical work of 180 hrs duration.

**Paper – I**  
**INORGANIC CHEMISTRY** **60 Hrs., Max Marks 33**

**UNIT-I**

**CHEMISTRY OF TRANSITION SERIES ELEMENTS**

Transition Elements: Position in periodic table, electronic configuration, General Characteristics, viz., atomic and ionic radii, variable oxidation states, ability to form complexes, formation of coloured ions, magnetic moment  $\mu_{so}$  (spin only) and  $\mu_{eff}$  and catalytic behaviour. General comparative treatment of 4d and 5d elements with their 3d analogues with respect to ionic radii, oxidation states and magnetic properties.

**UNIT-II**

**A. OXIDATION AND REDUCTION:** Redox potential, electrochemical series and its applications, Principles involved in extraction of the elements.

**B. COORDINATION COMPOUNDS:** Werner's theory and its experimental verification, IUPAC nomenclature of coordination compounds, isomerism in coordination compounds. Stereochemistry of complexes with 4 and 6 coordination numbers. Chelates, polynuclear complexes.

**UNIT-III**

**COORDINATION CHEMISTRY**

Valence bond theory (inner and outer orbital complexes), electroneutrality principle and back bonding. Crystal field theory, Crystal field splitting and stabilization energy, measurement of  $10 Dq$  ( $\Delta_o$ ), CFSE in weak and strong fields, pairing energies, factors affecting the magnitude of  $10 Dq$  ( $\Delta_o$ ,  $\Delta_t$ ). Octahedral vs. tetrahedral coordination.

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## UNIT-IV

### A. CHEMISTRY OF LANTHANIDE ELEMENTS

Electronic structure, oxidation states and ionic radii and lanthanide contraction, complex formation, occurrence and isolation, lanthanide compounds.

### B. CHEMISTRY OF ACTINIDES

General features and chemistry of actinides, chemistry of separation of Np, Pu and Am from uranium, similarities between the latter actinides and the latter lanthanides

## UNIT-V

**A. ACIDS BASES :** Arrhenius, Bronsted-Lowry, conjugate acids and bases, relative strengths of acids and bases, the Lux-flood, Solvent system and Lewis concepts of acids and bases.

### B. NON-AQUEOUS SOLVENTS

.Physical properties of a solvent, types of solvents and their general characteristics, reaction in non-aqueous solvents with reference to liquid ammonia and liquid sulphur dioxide, HF, H<sub>2</sub>SO<sub>4</sub> , Ionic liquids.

### REFERENCE BOOKS

1. Basic Inorganic Chemistry, F. A. Cotton, G. Wilkinson and P. L. Gaus, Wiley
2. Concise Inorganic Chemistry, J. D. Lee, ELBS
3. Concepts of Models of Inorganic Chemistry, B. Douglas, D. Mc Daniel and J. Alexander, John Wiley.
4. Inorganic Chemistry, D. E. Shriver, P. W. Atkins and C. H. Langford, Oxford.
5. Inorganic Chemistry, W. W. Porterfield, Addison – Wiley.
6. Inorganic Chemistry, A. G. Sharp, ELBS.
7. Inorganic Chemistry, G. L. Miessler and D. A. Tarr, Prentice Hall.
8. Advanced Inorganic Chemistry, Satya Prakash.
9. Advanced Inorganic Chemistry, Agarwal and Agarwal
10. Advanced Inorganic Chemistry, Puri, Sharma, S. Naginchand
11. Inorganic Chemistry, Madan, S. Chand
12. Aadhunik Akarbanic Rasayan, A. K. Shrivastav & P. C. Jain, Goel Pub
13. Uchchattar Akarbanic Rasayan, satya Prakash & G. D. Tuli, Shyamal Prakashan
14. Uchchattar Akarbanic Rasayan, Puri & Sharma
15. Selected topic in Inorganic Chemistry by Madan Malik & Tuli, S. Chand.

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**UNIT-I**

**CHEMISTRY OF ORGANIC HALIDES**

Alkyl halides: Methods of preparation, nucleophilic substitution reactions –  $S_N1$ ,  $S_N2$  and  $S_Ni$  mechanisms with stereochemical aspects and effect of solvent etc.; nucleophilic substitution, elimination reactions.

Aryl halides: Preparation, including preparation from diazonium salts, Nucleophilic Aromatic Substitution;  $S_NAr$ , Benzyne mechanism. Relative reactivity of alkyl, allyl/benzyl, vinyl and aryl halides towards nucleophilic substitution reactions.

**UNIT-II**

**ALCOHOLS**

- A. Alcohols: Nomenclature, preparation, properties and relative reactivity of  $1^\circ$ ,  $2^\circ$ ,  $3^\circ$  alcohols, Bouvaelt-Blanc Reduction for the preparation of alcohols, Dihydric alcohols – methods of formation, chemical reactions of vicinal glycols, oxidative cleavage [ $Pb(OAc)_4$  and  $HIO_4$ ] and pinacol-pinacolone rearrangement.
- B. Trihydric alcohols - Nomenclature, methods of formation, chemical reactions of glycerol.

**PHENOLS**

- A. Structure and bonding in phenols, physical properties and acidic character, Comparative acidic strength of alcohols and phenols, acylation and carboxylation.
- B. Mechanism of Fries rearrangement, Claisen rearrangement, Gatterman synthesis, Hauben-Hoesh reaction, Lederer-Manasse reaction and Reimer-Tiemann reaction.

**UNIT-III**

**ALDEHYDES AND KETONES**

- A. Nomenclature, structure and reactivity of carbonyl group. General methods of preparation of aldehydes and ketones.
- Mechanism of nucleophilic addition to carbonyl groups: Benzoin, Aldol, Perkin and Knoevenagel condensation. Condensation with ammonia and its derivatives, Wittig reaction, Mannich reaction, Beckmann and Benzil- Benzilic rearrangement.
- B. Use of acetate as protecting group, Oxidation of aldehydes, Baeyer-Villiger oxidation of ketones, Cannizzaro reaction, MPV, Clemmensen reduction, Wolf-Kishner reaction,  $LiAlH_4$  and  $NaBH_4$  reduction. Halogenation of enolizable ketones, An introduction to  $\alpha,\beta$ -unsaturated aldehydes and ketones.

B.Sc.-II

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## UNIT-IV

### A. CARBOXYLIC ACIDS

Preparation, Structure and bonding, Physical and chemical properties including, acidity of carboxylic acids, effects of substituents on acid strength, Hell-Volhard Zeilinsky reaction. Reduction of carboxylic groups, Mechanism of decarboxylation.

Di carboxylic acids: Methods of formation and effect of heat and dehydrating agents, Hydroxyacids.

### B. CARBOXYLIC ACID DERIVATIVES

Structure of acid chlorides, esters, amides and acid anhydrides, Relative stability of acyl derivatives.

Physical properties, inter-conversion of acid derivatives by nucleophilic acyl substitution.

Mechanism of acid and base catalyzed esterification and hydrolysis.

## UNIT-V

### ORGANIC COMPOUNDS OF NITROGEN

A. Preparation of nitroalkanes and nitroarenes. Chemical reactions of nitroalkanes. Mechanism of nucleophilic substitution in nitroarenes and their reduction in acidic, neutral and alkaline medium.

B. Reactivity, structure and nomenclature of amines, physical properties. Stereochemistry of amines. Separation of mixture of primary, secondary and tertiary amines. Structural features affecting basicity of amines. Preparation of alkyl and aryl amines (reduction of nitro compounds and nitriles), reductive amination of aldehydic and ketonic compounds. Gabriel-Phthalimide reaction, Hofmann-Bromamide reaction, Reactions of amines, electrophilic aromatic substitution of aryl amines, Reaction of amines with nitrous acid. Synthetic transformations of aryl diazonium salts, Azo coupling.

### REFERENCE BOOKS

1. Organic Chemistry, Morrison and Boyd, Prentice-Hall.
2. Organic Chemistry, L. G. Wade Jr. Prentice Hall.
3. Fundamentals of Organic Chemistry, Solomons, John Wiley.
4. Organic Chemistry, Vol I, II, III S. M. Mukherjee, S. P. Singh and R. P. Kapoor, Wiley Easters (New Age).
5. Organic Chemistry, F. A. Carey, McGraw Hill.
6. Introduction to Organic Chemistry, Struweiasser, Heathcock and Kosover, Macmillan.
7. Organic Chemistry, P. L. Soni.
8. Organic Chemistry, Bahl and Bahl.
9. Organic Chemistry, Joginder Singh.
10. Carbanic Rasayan, Bahl and Bahl.
11. Carbanic Rasayan, R. N. Singh, S. M. I. Gupta, M. M. Bakidia & S. K. Wadhwa.
12. Carbanic Rasayan, Joginder Singh.

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## UNIT-I

### A. THERMODYNAMICS-I

Intensive and extensive variables; state and path functions; isolated, closed and open systems; Zeroth law of thermodynamics. First law: Concept of heat, work, internal energy and statement of first law; enthalpy, Relation between heat capacities, calculations of  $q$ ,  $w$ ,  $U$  and  $H$  for reversible, irreversible and free expansion of gases under isothermal and adiabatic conditions. Joule-Thomson expansion, inversion temperature of gases, expansion of ideal gases under isothermal and adiabatic condition

### B. THERMO CHEMISTRY

Thermochemistry, Laws of Thermochemistry, Heats of reactions, standard states; enthalpy of formation of molecules and ions and enthalpy of combustion and its applications; calculation of bond energy, bond dissociation energy and resonance energy from thermochemical data, effect of temperature (Kirchhoff's equations) and pressure on enthalpy of reactions, Adiabatic flame temperature, explosion temperature.

## UNIT-II

### A. THERMODYNAMICS-II

Second Law of Thermodynamics: Spontaneous process, Second law, Statement of Carnot cycle and efficiency of heat engine, Carnot's theorem, thermodynamic state of temperature. Concept of entropy: Entropy change in a reversible and irreversible process, entropy change in isothermal reversible expansion of an ideal gas, entropy change in isothermal mixing of ideal gases, physical signification of entropy, Molecular and statistical interpretation of entropy.

B. Gibbs and Helmholtz free energy, variation of  $G$  and  $A$  with pressure, volume, temperature, Gibbs-Helmholtz equation, Maxwell relations, Elementary idea of Third law of Thermodynamics, concept of residual entropy, calculation of absolute entropy of molecule.

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## UNIT III

### A CHEMICAL EQUILIBRIUM

Criteria of thermodynamic equilibrium, degree of advancement of reaction, chemical equilibria in ideal gases. Concept of Fugacity, Thermodynamic derivation of relation between Gibbs free energy of reaction and reaction quotient. Coupling of exergonic and endergonic reactions. Equilibrium constants and their quantitative dependence on temperature, pressure and concentration. Thermodynamic derivation of relations between the various equilibrium constants  $K_p$ ,  $K_c$  and  $K_x$ . Le Chatelier principle (quantitative treatment). Equilibrium between ideal gas and a pure condensed phase.

### B IONIC EQUILIBRIA

Ionization of weak acids and bases, pH scale, common ion effect; dissociation constants of mono protic acids (exact treatment). Salt hydrolysis-calculation of hydrolysis constant, degree of hydrolysis and pH for different salts. Buffer solutions; derivation of Henderson equation and its applications. Solubility and solubility product of sparingly soluble salts – applications of solubility product principle.

## UNIT-IV

### PHASE EQUILIBRIUM

A. Phase rule, Phase, component and degree of freedom, derivation of Gibbs phase rule, Clausius-Claperon equation and its applications to Solid-Liquid, Liquid-Vapor and Solid-Vapor, limitation of phase rule, applications of phase rule to one component system: Water system and sulphur system.

Application of phase rule to two component system: Pb-Ag system, desilverization of lead, Zn-Mg system, Ferric chloride-water system, congruent and incongruent melting point and eutectic point.

Three component system: Solid solution liquid pairs.

B. Nernst distribution law, Henry's law, application, solvent extraction

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## UNIT V

### PHOTOCHEMISTRY

Characteristics of electromagnetic radiation, Interaction of radiation with matter, difference between thermal and photochemical processes, Lambert-Beer's law and its limitations, physical significance of absorption coefficients. Laws of photochemistry: Grothus-Draper law, Stark-Einstein law, quantum yield, actinometry, examples of low and high quantum yields, Photochemical equilibrium and the differential rate of photochemical reactions, Quenching, Role of photochemical reaction in biochemical process.

Jablonski diagram depicting various process occurring in the excited state, qualitative description of fluorescence, phosphorescence, non-radiative processes (internal conversion, intersystem crossing), photosensitized reactions, energy transfer processes {simple examples}, photostationary states, Chemiluminescence.

### REFERENCE BOOKS

1. Physical Chemistry, G. M. Barrow, International student edition, McGraw Hill.
2. University General Chemistry, C. N. R. Rao, Macmillan.
3. Physical Chemistry, R. A. Alberty, Wiley Eastern.
4. The elements of physical chemistry, Wiley Eastern.
5. Physical Chemistry through problems, S. K. Dogra & S. Dogra, Wiley Eastern.
6. Physical Chemistry, B. D. Khosla,.
7. Physical Chemistry, Puri & Sharma.
8. Bhautik Rasayan, Puri, Sharma and Pathania, Vishal Publishing Company.
9. Bhautik Rasayan, P. L. Soni.
10. Bhautik Rasayan, Bahl and Tuli.
11. Physical Chemistry, R. L. Kapoor, Vol I-IV .
12. Chemical kinetics, K. J. Laidler, Pearson Educations, New Delhi (2004).

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## LABORATORY COURSE

**INORGANIC CHEMISTRY**

Qualitative semimicro analysis of mixtures containing 5 radicals. Emphasis should be given to the understanding of the chemistry of different reactions. The following radicals are suggested:

$\text{CO}_3^{2-}$ ,  $\text{NO}_2^-$ ,  $\text{S}^{2-}$ ,  $\text{SO}_3^{2-}$ ,  $\text{S}_2\text{O}_3^{2-}$ ,  $\text{CH}_3\text{COO}^-$ ,  $\text{F}^-$ ,  $\text{Cl}^-$ ,  $\text{Br}^-$ ,  $\text{I}^-$ ,  $\text{NO}_3^-$ ,  $\text{BO}_3^{3-}$ ,  $\text{C}_2\text{O}_4^{2-}$ ,  $\text{PO}_4^{3-}$ ,  $\text{NH}_4^+$ ,  $\text{K}^+$ ,  $\text{Pb}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Bi}^{3+}$ ,  $\text{Sn}^{2+}$ ,  $\text{Sb}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Al}^{3+}$ ,  $\text{Cr}^{3+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Ba}^{2+}$ ,  $\text{Sr}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ .

Mixtures should preferably contain one interfering anion, or insoluble component ( $\text{BaSO}_4$ ,  $\text{SrSO}_4$ ,  $\text{PbSO}_4$ ,  $\text{CaF}_2$  or  $\text{Al}_2\text{O}_3$ ) or combination of anions e.g.  $\text{CO}_3^{2-}$  and  $\text{SO}_3^{2-}$ ,  $\text{NO}_2^-$  and  $\text{NO}_3^-$ ,  $\text{Cl}^-$ ,  $\text{Br}^-$ , and  $\text{I}^-$ .

**Volumetric analysis**

- Determination of acetic acid in commercial vinegar using NaOH.
  - Determination of alkali content-antacid tablet using HCl.
  - Estimation of calcium content in chalk as calcium oxalate by permanganometry.
  - Estimation of hardness of water by EDTA.
  - Estimation of ferrous & ferric by dichromate method.
  - Estimation of copper using thiosulphate.
- Principles involved in chromatographic separations. Paper chromatographic separation of following metal ions: i. Ni (II) and Co (II) ii. Fe (III) and Al (III)

**ORGANIC CHEMISTRY**

- Detection of elements (X, N, S).
- Qualitative analysis of unknown organic compounds containing simple functional groups (alcohols, carboxylic acids, phenols, nitro, amine, amide, and carbonyl compounds, carbohydrates)
- Preparation of Organic Compounds:
  - m-dinitrobenzene, (ii) Acetanilide, (iii) Bromo/Nitro-acetanilide, (iv) Oxidation of primary alcohols-Benzoic acid from benzylalcohol, (v) azo dye.

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## PHYSICAL CHEMISTRY

### Transition Temperature

- Determination of the transition temperature of the given substance by thermometric/ dilatometric method (e.g.  $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$  /  $\text{SrBr}_2 \cdot 2\text{H}_2\text{O}$ ).

### Thermochemistry

- Determination of heat capacity of a calorimeter for different volumes using change of enthalpy data of a known system (method of back calculation of heat capacity of calorimeter from known enthalpy of solution or enthalpy of neutralization).
- Determination of heat capacity of the calorimeter and enthalpy of neutralization of hydrochloric acid with sodium hydroxide.
- To determine the solubility of benzoic acid at different temperature and to determine  $\Delta H$  of the dissolution process.
- To determine the enthalpy of neutralization of a weak acid/ weak base versus strong base/ strong acid and determine the enthalpy of ionization of the weak acid/ weak base.
- To determine the enthalpy of solution of solid calcium chloride and calculate the lattice energy of calcium chloride from its enthalpy data using Born Haber cycle.

### Phase Equilibrium

- To study the effect of a solute (e.g. NaCl, Succinic acid) on the critical solution temperature of two partially miscible liquids (e.g. phenol-water system) and to determine the concentration of that solute in the given phenol-water system.
- To construct the phase diagram of two component system (e.g. diphenylamine– benzophenone) by cooling curve method.
- Distribution of acetic/ benzoic acid between water and cyclohexane.
- Study the equilibrium of at least one of the following reactions by the distribution method:
  - (i)  $\text{I}_2(\text{aq}) + \text{I}^- \rightarrow \text{I}_3(\text{aq})^{2-}$
  - (ii)  $\text{Cu}^{2+}(\text{aq}) + n\text{NH}_3 \rightarrow \text{Cu}(\text{NH}_3)_n$

### Molecular Weight Determination

Determination of molecular weight by Rast Camphor and Landsburger method.

**Note: Experiments may be added/ deleted subject to availability of time and facilities.**

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## Reference Books

1. Mann, F.G. & Saunders, B.C. Practical Organic Chemistry, Pearson Education (2009)
2. Furniss, B.S., Hannaford, A.J., Smith, P.W.G. & Tatchell, A.R. Practical Organic Chemistry, 5th Ed. Pearson (2012)
3. Ahluwalia, V.K. & Aggarwal, R. Comprehensive Practical Organic Chemistry: Preparation and Quantitative Analysis, University Press (2000). 22
4. Ahluwalia, V.K. & Dhingra, S. Comprehensive Practical Organic Chemistry: Qualitative Analysis, University Press (2000).
5. Khosla, B. D.; Garg, V. C. & Gulati, A. Senior Practical Physical Chemistry, R. Chand & Co.: New Delhi (2011). Garland, C. W.; Nibler, J. W. & Shoemaker, D. P. Experiments in Physical Chemistry 8th Ed.; McGraw-Hill: New York (2003).
6. Halpern, A. M. & McBane, G. C. Experimental Physical Chemistry 3rd Ed.; W.H. Freeman & Co.: New York

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Three Experiments are to be performed.

1. Inorganic – Qualitative semimicro analysis of mixtures. **12 marks**

OR

One experiment from synthesis and analysis by preparing the standard solution.

2. (a) Identification of the given organic compound & determine its M.Pt./B.Pt.

**6 marks**

(b) Determination of Rf value and identification of organic compounds by paper chromatography.

**6 marks**

3. Any one physical experiment that can be completed in two hours including calculations.

**12 marks**

4. Viva

**10 marks**

5. Sessional

**04 marks**

In case of Ex-Students one marks will be added to each of the experiment.

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## THERMODYNAMICS, KINETIC THEORY AND STATISTICAL PHYSICS

- Unit-1** The laws of thermodynamics : The Zeroth law, first law of thermodynamics, internal energy as a state function, reversible and irreversible change, Carnot's cycle, Carnot theorem, second law of thermodynamics. Clausius theorem inequality. Entropy, Change of entropy in simple cases (i) Isothermal expansion of an ideal gas (ii) Reversible isochoric process (iii) Free adiabatic expansion of an ideal gas. Concept of entropy, Entropy of the universe. Entropy change in reversible and irreversible processes, Entropy of Ideal gas, Entropy as a thermodynamic variable, S-T diagram, Principle of increase of entropy. The thermodynamic scale of temperature, Third law of thermodynamics, Concept of negative temperature.
- Unit-2** Thermodynamic functions, Internal energy, Enthalpy, Helmholtz function and Gibb's free energy, Maxwell's thermodynamical equations and their applications, TdS equations, Energy and heat capacity equations Application of Maxwell's equation in Joule-Thomson cooling, adiabatic cooling of a system, Van der Waals gas, Clausius-Clapeyron heat equation. Blackbody spectrum, Stefan-Boltzmann law, Wien's displacement law, Rayleigh-Jean's law, Planck's quantum theory of radiation.
- Unit-3** Maxwellian distribution of speeds in an ideal gas: Distribution of speeds and velocities, experimental verification, distinction between mean, rms and most probable speed values. Doppler broadening of spectral lines. Transport phenomena in gases: Molecular collisions mean free path and collision cross sections. Estimates of molecular diameter and mean free path. Transport of mass, momentum and energy and interrelationship, dependence on temperature and pressure.  
Behaviour of Real Gases: Deviations from the Ideal Gas Equation. The Virial Equation. Andrew's Experiments on CO<sub>2</sub> Gas. Critical Constants.
- Unit-4** The statistical basis of thermodynamics: Probability and thermodynamic probability, principle of equal a priori probabilities, statistical postulates. Concept of Gibb's ensemble, accessible and inaccessible states. Concept of phase space,  $\gamma$  phase space and  $\mu$  phase space. Equilibrium before two systems in thermal contact, probability and entropy, Boltzmann entropy relation. Boltzmann canonical distribution law and its applications, law of equipartition of energy.


  
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21. Study of lissajous figures using CRO.
22. Study of VTVM.
23. Study of RC and TC coupled amplifiers.
24. Study of AF and RF oscillators.
25. Find roots of  $f(x) = 0$  by using Newton-Raphson Method.
26. Find root of  $f(x) = 0$  by using secant method.
27. Integration by Simpson rule.
28. To find the value of V at
29. String manipulations.
30. Towers of Hanoi (Non-recursive).
31. Finding first four perfect numbers.
32. Quadratic interpolation using Newton's forward-difference formula of degree two.

**TEXT AND REFERENCE BOOKS:**

1. B.G. Strechman, Solid state electronics devices II edition (Prentice-Hall of India New Delhi 1986)
2. W.D. Stanley, Electronics devices, circuits and applications (Prentice-Hall new jersey, USA 1988).
3. S. Lipschutz and A Poe; Schaum's outline of theory and problems of programming with Fortran (Mc Graw-Hill Book Co. Singapore, 1986).
4. C Dixon, Numerical Analysis.

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*P. S. S.*

## TEXT AND REFERENCE BOOKS:

1. Introduction to solid state physics: C. Kittel.
2. Solid State Physics: A.J. Dekkar.
3. Electronic Circuits: Mottershead.
4. Electronic Circuits: Millman and Halkias.
5. Semiconductor Devices: S.M. Sze.
6. Electronic devices: T.L. Floyd.
7. Device and Circuits: J. Millman and C. Halkias.
8. Electronic Fundamental and Applications: D. Chatopadhyay and P.C. Rakshit.
9. Electricity and Magnetism: K.K. Tiwari.

## PRACTICALS

Minimum 16 (Eight from each group)

Experiments out of the following or similar experiments of equal standard

1. Determination of Planck's constant.
2. Determination of  $e/m$  by using Thomson tube.
3. Determination of  $e$  by Millikan's methods.
4. Study of spectra of hydrogen and deuterium (Rydberg constant and ratio of masses of electron proton).
5. Absorption spectrum of iodine vapour.
6. Study of alkali or alkaline earth spectra using a concave grating.
7. Study of Zeeman effect for determination of a Lande  $g$ -factor.
8. Analysis of a given band spectrum.
9. Study of Raman spectrum using laser as an excitation source.
10. Study of absorption of alpha and beta rays.
11. Study of statistics in radioactive measurement.
12. Coniometric study of crystal faces.
13. Determination of dielectric constant.
14. Hysteresis curve of transformer core.
15. Hall-probe method for measurement of magnetic field.
16. Specific resistance and energy gap of semiconductor.
17. Characteristics of transistor.
18. Characteristics of tunnel diode.
19. Study of voltage regulation system.
20. Study of regulated power supply.

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Session 2019-20

PHYSICS

Paper-II

SOLID STATE PHYSICS, SOLID STATE DEVICES AND ELECTRONICS

- Unit-1** Amorphous and crystalline solids, Elements of symmetry, seven crystal system, Cubic lattices, Crystal planes, Miller indices, Laue's equation for X-ray diffraction, Bragg's Law, Bonding in solids, classification. Cohesive energy of solid, Madelung constant, evaluation of Parameters, Specific heat of solids, classical theory (Dulong-Petit's law), Einstein and Debye theories, Vibrational modes of one dimensional monoatomic lattice, Dispersion relation, Brillouin Zone.
- Unit-2** Free electron model of a metal, Solution of one dimensional Schrödinger equation in a constant potential, Density of states, Fermi Energy, Energy bands in a solid (Kronig-Penny model without mathematical details), Difference between Metals, Insulator and Semiconductors, Hall effect, Dia, Para and Ferromagnetism, Langevin's theory of dia and para-magnetism, Curie- Weiss's Law, Qualitative description of Ferromagnetism (Magnetic domains), B-H curve and Hysteresis loss.
- Unit-3** Intrinsic and extrinsic semi conductors, Concept of Fermi level, Generation and recombination of electron hole pairs in semiconductors, Mobility of electrons and holes, drift and diffusion currents, p-n junction diode, depletion width and potential barrier, junction capacitance, I-V characteristics, Tunnel diode, Zener diode, Light emitting diode, solar cell, Bipolar transistors, pnp and npn transistors, characteristics of transistors, different configurations, current amplification factor, FET and MOSFET Characteristics.
- Unit-4** Half and full wave rectifier, rectifier efficiency ripple factor, Bridge rectifier, Filters, Inductor filter, L and  $\pi$  section filters, Zener diode, regulated power supply using zener diode, Applications of transistors, Bipolar Transistor as amplifier, h-parameter, h-parameter equivalent circuit, Transistor as power amplifier, Transistor as oscillator, principle of an oscillator and Bark Hausen's condition, requirements of an oscillator, Wein-Bridge oscillator and Hartley oscillator.
- Unit-5** Digital Circuits: Difference between Analog and Digital Circuits, Binary Numbers, Decimal to Binary and Binary to Decimal Conversion, AND, OR and NOT Gates (Realization using Diodes and Transistor), NAND and NOR Gates as Universal Gates, XOR and XNOR Gate, De Morgan's Theorems, Boolean Laws, Simplification of Logic Circuit using Boolean Algebra, Digital to Analog Converter, Analog to Digital Converter.

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**Unit-5** Structure of nuclei:- Basic Properties of Nuclei: (1) Mass, (2) Radii, (3) Charge, (4) Angular Momentum, (5) Spin, (6) Magnetic Moment ( $\mu$ ), (7) Stability and (8) Binding Energy, Nuclear Models:- Liquid Drop Model, Mass formula, Shell Model, Types of Nuclear reactions, laws of conservation, Q-value of reactions, Interaction of Energetic particles with matter, Ionization chamber, GM Counter, Cloud Chambers, Fundamental Interactions, Classification of Elementary Particles, Particles and Antiparticles, Baryons, Hyperons, Leptons, and Mesons, Elementary Particle Quantum Numbers: Baryon Number, Lepton Number, Strangeness, Electric Charge, Hypercharge and Isospin, introductory idea of discovery of Higg's Boson.

#### TEXT AND REFERENCE BOOKS:

1. H.S. Mani and G.K. Metha: "Introduction to Modern Physics" (Affiliated East-West Press, 1989).
2. A Beiser, "Prospective of Modern Physics".
3. H.E. White, "Introduction to Atomic Physic".
4. Barrow, "Introduction to Molecular Physics".
5. R.P. Feynman, R.B. Leighton and M Sands, "The Feynman Lectures on Physics", Vol.III (B.I. Publications, Bombay, Delhi, Calcutta, Madras).
6. T.A. Littlefield and N Thorley, "Atomic and Nuclear Physics" (Engineering Language Book Society)
7. H.A. Enge, "Introduction to Nuclear Physics", (Addision-Wesly)
8. Eisenberg and Resnick, "Quantum Physics of Atoms, Molecules, Solids, Nuclei and Particles" (John Wiley)
9. D.P. Khandelwal, "Optics and Atomic Physics", (Himalaya Publishing House, Bombay, 1988).
10. Quarks and Leptons, F. Halzen and A.D. Martin, Wiley India, New Delhi, 1984.
11. Radiation detection and measurement, G.F. Knoll (John Wiley & Sons, 2000).
12. Theoretical Nuclear Physics, J.M. Blatt & V.F. Weisskopf (Dover Pub.Inc., 1991).

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A small signature "S.P." below the initials.

## MATHEMATICS

There shall be three compulsory papers. Each paper of 50 marks is divided into five units and each unit carry equal marks.

### B.Sc. Part-II

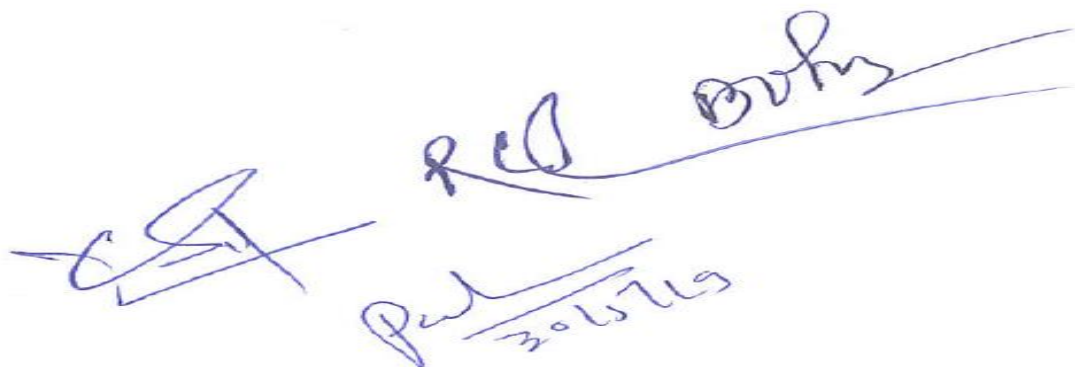
#### Paper-I

#### ADVANCED CALCULUS

- UNIT-I Definition of a sequence. Theorems on limits of sequences. Bounded and monotonic sequences. Cauchy's convergence criterion. Series of non-negative terms. Comparison tests, Cauchy's integral test, Ratio tests, Raabe's, Logarithmic, De Morgan and Bertrand's tests. Alternating series, Leibnitz's theorem. Absolute and conditional convergence.
- UNIT-II Continuity, Sequential continuity, Properties of continuous functions, Uniform continuity, Chain rule of differentiability, Mean value theorems and their geometrical interpretations. Darboux's intermediate value theorem for derivatives, Taylor's theorem with various forms of remainders.
- UNIT-III Limit and continuity of functions of two variables. Partial differentiation. Change of variables. Euler's theorem on homogeneous functions. Taylor's theorem for functions of two variables. Jacobians.
- UNIT-IV Envelopes, evolutes. Maxima, minima and saddle points of functions of two variables. Lagrange's multiplier method.
- UNIT-V Beta and Gamma functions, Double and triple integrals, Dirichlet's integrals, Change of order of integration in double integrals.

#### REFERENCES :

1. Gabriel Klaumber, Mathematical Analysis, Marcel Dekkar, Inc. New York, 1975.
2. T.M. Apostol, Mathematical Analysis, Narosa Publishing House, New Delhi, 1985.
3. R.R. Goldberg, Real Analysis, Oxford & I.B.H. Publishing Co., New Delhi, 1970.
4. D. Soma Sundaram and B. Choudhary, A First Course in Mathematical Analysis, Narosa Publishing House, New Delhi, 1997.
5. P.K. Jain and S.K. Kaushik, An introduction to Real Analysis, S. Chand & Co., New Delhi, 2000.
6. Gorakh Prasad, Differential Calculus, Pothishala Pvt. Ltd., Allahabad.
7. Murray R. Spiegel, Theory and Problems of Advanced Calculus, Schaum Publishing Co., New York.
8. Gorakh Prasad, Integral Calculus, Pothishala Pvt. Ltd., Allahabad.
9. S.C. Malik, Mathematical Analysis, Wiley Eastern Ltd., New Delhi.
10. O.E. Stanaitis, An Introduction to Sequences, Series and Improper Integrals, Holden-Dey, Inc., San Francisco, California.
11. Earl D. Rainville, Infinite Series, The Macmillan Company, New York.
12. Chandrika Prasad, Text Book on Algebra and Theory of Equations, Pothishala Pvt. Ltd., Allahabad.
13. N. Piskunov, Differential and Integral Calculus, Peace Publishers, Moscow.
14. Shanti Narayan, A Course of Mathematical Analysis, S.Chand and Company, New Delhi.

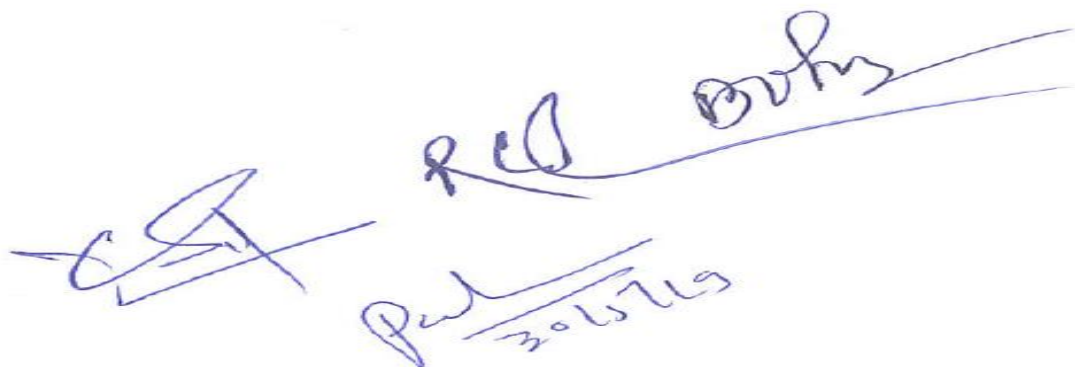
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**B.Sc. Part-II**  
**Paper-II**  
**DIFFERENTIAL EQUATIONS**

- UNIT-I Series solutions of differential equations- Power series method, Bessel and Legendre functions and their properties-convergence, recurrence and generating relations, Orthogonality of functions, Sturm-Liouville problem, Orthogonality of eigen-functions, Reality of eigen values, Orthogonality of Bessel functions and Legendre polynomials.
- UNIT-II Laplace Transformation- Linearity of the Laplace transformation, Existence theorem for Laplace transforms, Laplace transforms of derivatives and integrals, Shifting theorems. Differentiation and integration of transforms. Convolution theorem. Solution of integral equations and systems of differential equations using the Laplace transformation.
- UNIT-III Partial differential equations of the first order. Lagrange's solution, Some special types of equations which can be solved easily by methods other than the general method, Charpit's general method of solution.
- UNIT-IV Partial differential equations of second and higher orders, Classification of linear partial differential equations of second order, Homogeneous and non-homogeneous equations with constant coefficients, Partial differential equations reducible to equations with constant coefficients, Monge's methods.
- UNIT-V Calculus of Variations- Variational problems with fixed boundaries- Euler's equation for functionals containing first order derivative and one independent variable, Extremals, Functionals dependent on higher order derivatives, Functionals dependent on more than one independent variable, Variational problems in parametric form, invariance of Euler's equation under coordinates transformation.
- Variational Problems with Moving Boundaries- Functionals dependent on one and two functions, One sided variations.
- Sufficient conditions for an Extremum- Jacobi and Legendre conditions, Second Variation. Variational principle of least action.

REFERENCES :

1. Erwin Kreyszig, Advanced Engineering Mathematics, John Wiley & Sons, Inc., New York, 1999.
2. D.A. Murray, Introductory Course on Differential Equations, Orient Longman, (India), 1967.
3. A.R. Forsyth, A Treatise on Differential Equations, Macmillan and Co. Ltd., London.
4. Lan N. Sneddon, Elements of Partial Differential Equations, McGraw-Hill Book Company, 1988.
5. Francis B. Hilderbrand, Advanced Calculus for Applications, Prentice Hall of India Pvt. Ltd., New Delhi, 1977.
6. Jane Cronin, Differential equations, Marcel Dekkar, 1994.
7. Frank Ayres, Theory and Problems of Differential Equations, McGraw-Hill Book Company, 1972.
8. Richard Bronson, Theory and Problems of Differential Equations, McGraw-Hill, Inc., 1973.
9. A.S. Gupta, Calculus of variations with-Applications, Prentice-Hall of India, 1997.
10. R. Courant and D. Hilbert, Methods of Mathematical Physics, Vols. I & II, Wiley-Interscience, 1953.
11. I.M. Gelfand and S.V. Fomin, Calculus of Variations, Prentice-Hill, Englewood Cliffs (New Jersey), 1963.
12. A.M. Arthurs, Complementary Variational Principles, Clarendon Press, Oxford, 1970.
13. V. Kornkov, Variational Principles of Continuum Mechanics with Engineering Applications, Vol. I, Reidel Publ. : Dordrecht, Holland, 1985.
14. T. Oden and J.N. Reddy, Variational Methods in Theoretical Mechanics, Springer-Verlag, 1976.

  
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**B.Sc. Part-II**  
**Paper-III**  
**MECHANICS**

**STATICS**

UNIT-I Analytical conditions of Equilibrium, Stable and unstable equilibrium. Virtual work, Catenary.

UNIT-II Forces in three dimensions, Poinsot's central axis, Null lines and planes.

**DYNAMICS**

UNIT-III Simple harmonic motion. Elastic strings. Velocities and accelerations along radial and transverse directions, Projectile, Central orbits.

UNIT-IV Kepler's laws of motion, velocities and acceleration in tangential and normal directions, motion on smooth and rough plane curves.

UNIT-V Motion in a resisting medium, motion of particles of varying mass, motion of a particle in three dimensions, acceleration in terms of different co-ordinate systems.

**REFERENCES :**

1. S.L. Loney, Statics, Macmillan and Company, London.
2. R.S. Verma, A Text Book on Statics, Pothishala Pvt. Ltd., Allahabad.
3. S.L. Loney, An Elementary Treatise on the Dynamics of a particle and of rigid bodies, Cambridge University Press, 1956.

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*20/5/19*

## B.Sc.–II (BOTANY) PAPER-I

### (PLANT TAXONOMY, ECONOMIC BOTANY, PLANT ANATOMY AND EMBRYOLOGY)

#### UNIT-I

Bentham and Hooker system of classification. Binomial Nomenclature, International Code of Nomenclature for Algae, Fungi, and plants (IUCN), Typification, numerical Taxonomy and chemotaxonomy. Preservation of Plant material and Herbarium techniques. Important botanical gardens and herbaria of India, Kew Botanical garden, England.

#### UNIT-II

Systematic position, distinguishing characters and economic importance of the following families, Ranunculaceae, Magnoliaceae, Brassicaceae, Rosaceae, Papaveraceae, Caryophyllaceae, Rutaceae, Cucurbitaceae, Apiaceae, Rubiaceae, Apocynaceae, Asclepiadaceae, Solanaceae, Malvaceae, Convolvulaceae, Orchidaceae, Acanthaceae, verbenaceae, Lamiaceae, Asteraceae, Fabaceae, Euphorbiaceae, Poaceae and Liliaceae.

#### UNIT-III

Economic Botany: Botanical name, family, part used and uses of the following economically important plants, fiber yielding plants; Cotton, jute, sun, hemp, coir. Timber yielding plants: Sal, Teak, Shisham and Pine. Medicinal plants: Kalmegh, Ashwagandha, Ghritkumari, Giloy, Brahmi, sarpandha, ---of medicinal plants of C.G. Food plants: Pearl millet, Buck of wheat, Sorghum, Soyabean, gram, Ground nut, Sugarcane and Potato. Fruit plants: Pear, Peach, Litchi. Spices: Cinnamon, Turmeric, Ginger, Asafoetida and Cumin. Beverages : Tea, Coffee Rubber Cultivation of important flowers: Chrysanthemum, Dahelia, Biodiesel plants Jatropha, Pongamia Ethnobotany in context of Chhattisgarh.

#### UNIT-IV

Plant Anatomy: Root and shoot apical meristems theories of root and shoot apex organization, permanent tissues, anatomy of root, stem and leaf of dicot and monocot, secondary growth in root and stem, Anatomical anomalies in the primary structure of stems (Nyctanthes, Boerhaavia, Casuarina), Anamolous secondary growth in Dracaena, Bignonia, Laptadenia.

#### UNIT-V

Embryology: Flower as a reproductive organ, anther, microsporogenesis, types of ovules, megasporogenesis, development of male and female gametophyte, pollination, mechanisms, self incompatibility, fertilization, endosperm, embryo, polyembryonoy, apomixes and parthenocarpy.

#### Books Recommended:

Amal  
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Rashmi  
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Dr. Anshu  
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Shruti  
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Singh, Pandey, Jain, *Diversity and Systematics of Seed Plants*, Rastogi Publications Merrut

Sharma OP, *Plant Taxonomy*, Tata Mc Graw Hill, New Delhi

Pandey BP, *Taxonomy of Angiosperms*, S. Chand Publishing, New Delhi

Pandey, BP, *Plant Anatomy*, S.Chand Publishing, New Delhi

Pandey, BP, *Economic Botany*, S.Chand Publishing, New Delhi

Bhojwani, SS and Bhatanagar SP, *Embryology of Angiosperm*, Vikas Publication House, New Delhi

Singh, Pandey, Jain, *Embryology of Angiosperms*, Rastogi Publication, Meerut

Sharma, V, Alum, A. *Ethnobotany*, Rastogi Publications, Meerut

Tayal, MS *Plant Anatomy*, Rastogi Publication, Meerut

(Dr. J.N. Verma)

Proff. & Head

Govt. D.B. Girls PG College

Raipur, (C.G.)

(Dr. Rekha Pimpalgaonkar)

Proff. & Head

Govt. N PG Science College

Raipur, (C.G.)

(Dr. Ranjana Shrivastava)

Proff. & Head

Govt. VYTPG Science College

Raipur, (C.G.)

(Mrs. Sanchal Moghe)

Govt. Bilasa Girls College, Bilaspur

(Mr. Shivakant Mishra)

(Mr. Sudheer Tiwari)

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**B.Sc.-II (BOTANY) PAPER-II**  
**(ECOLOGY AND PLANT PHYSIOLOGY)**

**UNIT-I**

Introduction and scope of ecology, environmental and ecological factors, Soil formation and soil profile, Liebig's law of minimum, Shelford's law of tolerance, morphological and anatomical adaptations in hydrophytes, xerophytes and epiphytes.

**UNIT-II**

Population and community characteristics, Raunkiaer's life forms, population interactions (e.g. Symbiosis, Amensalism etc.), succession, ecotone and edge effect, ecological niches, ecotypes, ecads, keystone species

Concept of ecosystem, trophic levels, flow of energy in ecosystem, food chain and food web, concept of ecological pyramids

Biogeochemical cycles: carbon cycle, nitrogen cycle and phosphorus cycle

**UNIT-III**

Plant water relations: Diffusion, permeability, osmosis, imbibitions, plasmolysis, osmotic potential and water potential, Types of soil water, water holding capacity, wilting, Absorption of water, theories of Ascent of sap, Mineral nutrition and absorption, Deficiency symptoms, Transpiration, stomatal movement, significance of transpiration, Factors affecting transpiration, guttation.

**UNIT-IV**

Photosynthesis: Photosynthetic apparatus and pigments, light reaction mechanism of ATP synthesis. C<sub>3</sub>, C<sub>4</sub> CAM pathway of carbon reduction, photorespiration, factors affecting photosynthesis.

Respiration: Aerobic and anaerobic respiration, Glycolysis, Krebs's cycle, factors affecting respiration, R.Q.

**UNIT-V**

Plant growth hormones: Auxin, Gibberellin, Cytokinin, Ethylene and Abscissic acid. Physiology of flowering, Florigen concept, Photoperiodism and Vernalization. Seed dormancy and germination, plant movement.

**Books Recommended:**

Koromondy, EJ. *Concepts of Ecology*, Prentice Hall, USA

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Singh, JS Singh SP and Gupta SR. *Ecology and Environmental Science and Conservation*, S. Chand Publishing, New Delhi

Sharma, PD. *Ecology and Environment*, Rastogi Publications, Merrut

Hopkins, WG and Huner, PA. *Introduction to Plant Physiology*, John Wiley and Sons.

Pandey SN and Sinha BK, *Plant Physiology*, Vikas Publishing, New Delhi

Taiz, L and Zeiger. E. *Plant Physiology*, 5<sup>th</sup> edition, Sinauer Associates Inc. M.A, USA

Srivastava, HS *Plant Physiology and Biotechnology*, Rastogi Publications, Meerut

### B.Sc. II (BOTANY)

#### Practical

1. Taxonomy: Detailed description and identification of locally available plants of the families as prescribed in the theory paper.
2. Economic Botany: Identification and comment on the plants and plant products belonging to different economic use categories
3. Preparation of Herbarium of local wild plants.
4. Quantitative vegetation analysis of a grassland ecosystem.
5. Anatomical characteristics of hydrophytes and xerophytes.
6. Demonstration of root pressure.
7. Demonstration of transpiration.
8. Demonstration of evolution of O<sub>2</sub> in photosynthesis, factors affecting of photosynthesis.
9. Comparison of R.Q. of different respiratory substrates.
10. Demonstration of fermentation.
11. Determination of BOD of a water body.
12. Demonstration of mitosis.

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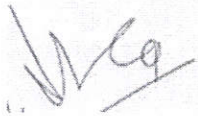
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## PRACTICAL SCHEME

TIME: 4 Hrs.

M.M. : 50

1.	Anatomy	08
2.	Economic Botany	04
3.	Physiology	08
4.	Ecology	10
5.	Spotting	10
6.	Viva-Voce	05
7.	Project Work/ Field Study	10

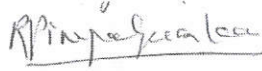


(Dr. J.N. Verma)

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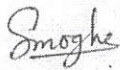


(Dr. Ranjana Shrivastava)

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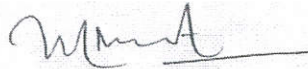
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Govt. Bilasa Girls College, Bilaspur



(Mr. Shivakant Mishra)

(Mr. Sudheer Tiwari)

  
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**Hemchand Yadav Vishwavidyala, Durg (C.G.)**  
**Zoology**  
**B.Sc. Part – II (2019-20)**  
**Paper – I**  
**(Anatomy and Physiology)**

Comparative Anatomy of various organ systems of vertebrates:

**Unit: I**

- Integument and its derivatives: structure of scales, hair and feathers
- Alimentary canal and digestive glands in vertebrates
- Respiratory organs : Gills and lung , air-sac in birds

**Unit: II**

- Endoskeleton: (a) Axial Skeleton- Skull and Vertebrae, (b) Appendicular Skeleton  
Limbs and girdles
- Circulatory System: Evolution of heart and aortic arches
- Urinogenital System: Kidney and excretory ducts

**Unit: III**

- Nervous System: General plan of brain and spinal cord
- Ear and Eye: structure and function
- Gonads and genital ducts

**Unit: IV**

- Digestion and absorption of dietary components
- Physiology of heart, cardiac cycle and ECG
- Blood Coagulation
- Respiration: mechanism and control of breathing

**Unit: V**

- Excretion: Physiology of excretion, osmoregulation
- Physiology of muscle contraction
- Physiology of nerve impulse, Synaptic transmission

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**Zoology**  
**B.Sc. Part – II (2019-20)**

Paper-II

VERTEBRATE ENDOCRINOLOGY, REPRODUCTIVE BIOLOGY  
BEHAVIOUR, EVOLUTION AND APPLIED ZOOLOGY

**Unit: I**

- Structure and function of Endocrine glands
- Hormone receptor
- Biosynthesis and secretion of thyroid, adrenal, ovarian and testicular hormones
- Endocrine disorder of pituitary, thyroid, adrenal and pancreas

**Unit:II**

- Reproductive cycle in vertebrates
- Menstruation, lactation and pregnancy
- Mechanism of parturition
- Hormonal regulation of gametogenesis

**Unit: III**

- Evidences of organic evolution.
- Theories of organic evolution.
- Variation, Mutation, Isolation and Natural selection.
- Evolution of Horse

**Unit:IV**

- Introduction to Ethology: Branches and concept of ethology.
- Patterns of Behaviour, Taxes, Reflexes, Drives and Stereotyped behaviour.
- Reproductive behavioural patterns.
- Drugs and behavior, Hormones and behaviour

**Unit:V**

- Prawn Culture
- Sericulture
- Apiculture
- Pisciculture
- Poultry keeping
- Elements of Pest Control: Chemical & Biological Control

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**Zoology**  
**B.Sc. Part II (2019-20)**  
**Practical**

The practical work in general shall be based on the syllabus prescribed and the students will be required to show the knowledge of the following:

- Study of the representative examples of the different chordates (Classified characters).
- Dissection of various systems of scoliodon-Afferent and Efferent branchial cranial nerves, internal ear.

**Alternative methods: By Clay/Thermacol/ Drawing/ Model etc.)**

- Simple microscopic technique through unstained or stained permanent mount.
- Study of prepared slides histological, as per theory papers.
- Study of limb girdles and vertebrae of Frog, Varanus, Fowl and Rabbit.
- Identification of species and individual of honey bee.
- Life cycle of honey bee and silkworm.
- Exercise based on Evolution and Animal behavior.

**Scheme of Practical Exam**

**Time: 3:30hrs**

- |   |    |
|---|----|
| • Major dissection (Cranial nerves/efferent branchial vessel) | 10 |
| • Exercise based on evolution                                 | 05 |
| • Exercise based on applied zoology                           | 05 |
| • Exercise based on animal behavior                           | 04 |
| • Spotting-8 (slides-4,bones-2,specimen-2)                    | 16 |
| • Viva  | 05 |
| • Sessional marks.  | 05 |

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# MICROBIOLOGY

## BSc-2<sup>nd</sup>

### Paper- I: Molecular Biology and Genetic Engineering

#### UNIT-1: FUNDAMENTALS OF MOLECULAR BIOLOGY

History and scope of molecular biology, concept and mechanism of heredity. DNA as genetic material- experimental evidences. DNA replication- mechanism, process and enzymes/proteins involved in replication.

#### UNIT-2: CENTRAL DOGMA OF PROTEIN SYNTHESIS

Transcription- initiation, elongation, termination, RNA polymerases and sigma factor. Transcription inhibitors (antibiotics, drugs). Translation- initiation, elongation and termination. Factors involved in translation. Genetic code.

#### UNIT-3: MUTATION AND DNA REPAIR MECHANISM

Introduction and Types of Gene mutations- Base substitution, frame shift mutation (insertion, deletion, miss-sense, nonsense mutation.) mutagens – physical and chemical. Reverse mutation in bacteria. DNA repair mechanism (mismatch repair, photo-reactivation, excision and SOS repair). Beneficial and harmful effect of mutation.

#### UNIT-4: GENE REGULATION

Concept of gene- Cistron, Recon, Muton. Operon Concept- lac Operon, tryptophan Operon, His Operon. Activator, Co-activator and Repressor. Introduction to Bioinformatics- Elementary genome Database.

#### UNIT-5: GENETIC ENGINEERING

Basic concept of Genetic Engineering, DNA modifying enzymes Restriction endonuclease, DNA ligase, terminal transferase. Vectors- pBR322, pUC19, BAC and YAC. Phage based vectors, expression of vector. Transformation – physical and chemical method. Bacterial Host. Screening of recombinant vector Blue white Screening, Colony Hybridization.

### Text Books Recommended:

1. Gene Cloning by T.A. Brown.
2. General Microbiology by Power and Daganiwala.
3. Zinssers Microbiology by KJ Wolfgang, McGraw- Hill Company.
4. Microbial Genetics by RM Stanley, F David and EC John.
5. Bacteriological Techniques by FJ Baker.
6. Molecular Biology of the Cell; 3<sup>rd</sup> Edition; Bruce Alberts ,et.al; Garland Publishing.
7. Cell biology; C.B. Powar; Himalaya Publishing House; Fifth edition
8. Cell & Molecular Biology; Gerald Karp; Fourth edition
9. A Textbook of Microbiology; Dubey&Maheshwari; S.chand& Sons.
10. Cell biology & Genetics; P. K. Gupta
11. Introduction to Bioinformatics; T K Atwood and D J Parry-Smith; Pearson Education Ltd

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**Paper- II: Bioinstrumentation and Biostatistics****UNIT-1: MICROSCOPY AND CENTRIFUGATION**

Simple and compound light microscope, Bright field, Dark field, Phase contrast and Electron microscope. Centrifugation- principle and types of centrifuges (analytical and preparatory), types of centrifugation- differential and rate zonal centrifugation.

**UNIT-2: pH metry and chromatography**

Principle of pH meter, types of electrodes, factors affecting pH measurements, and application of pH meter. Chromatography- principle, types- paper, TLC and column chromatography, HPLC.

**UNIT-3: SPECTROPHOTOMETRY**

Electromagnetic spectrum, Beers-Lamberts law, Types (Principles, working and application)- colorimeter, UV - Vis Spectrophotometry and IR- Spectrophotometry, Turbidometry.

**UNIT-4: Electrophoresis and X-Ray Diffraction**

Principle of electrophoresis, instrumentation and Application, types of Paper, Gel electrophoresis and Immunoelectrophoresis. X-ray diffraction- principle and application.

**UNIT-5: Biostatistics**

Data- Types, characteristics, presentation and distribution. Data analysis- central tendency (Mean, Median and Mode), Deviation (variance SD and SE). Concept of probability.

***Text Books Recommended:***

1. Introduction to Instrumental analysis by Robert Braun.
2. Instrumental Techniques by Upadhyay and Upadhyay.
3. Instrumental Methods of Chemical Analysis by BK Sharma.
4. Bio statistics; Sunder Rao
5. Statistical Methods; S. P. Gupta; Sultan Chand & Sons

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## PRACTICAL

M. M. 50

Determination of antibiotic resistance by plating method.  
 Assaying of microbial enzymes; Catalase, Proteases, Peroxidases,  
 Cellulase, Cellobioases, Amylase, Diastase.  
 Exercise on paper, thin layer, column chromatography.  
 Exercise on paper and gel electrophoresis.  
 determination of pH of various water and soil sample.  
 testing of lambert beer's law.  
 Determination of lamda max of dye by spectrophotometer  
 Isolation of resistant bacteria from soil and water sample

## Scheme of Practical Examination

Time - 4 hours

M.M. 50

1. Exercise on spectrophotometer/ pH meter	10
2. Exercise on chromatography	10
3. Exercise on genetics	05
4. Spotting (1-5)	10
5. Viva-Voce	05
6. Sessional	10

Total 50

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# HEMCHAND YADAV VISHWAVIDYALAYA, DURG (C.G.)

## Scheme of Examination

कक्षा	प्रश्नपत्र	विषय समूह	सैद्धा. अंक	प्रायो. अंक	योग
BSc. I year	I	भूगतिकी एवं भू-आकृति विज्ञान (Geodynamics & Geomorphology)	50	50	150
	II	खनिज एवं क्रिस्टल विज्ञान (Mineralogy & Crystallography)	50		
BSc. II year	I	शैलिकी (Petrology)	50	50	150
	II	संरचनात्मक भूविज्ञान (Structural Geology)	50		
BSc. III year	I	जीवाश्म विज्ञान एवं संस्तर विज्ञान (Palaeontology & Stratigraphy)	50	50	150
	II	भूसंसाधन एवं व्यावहारिक भूविज्ञान (Earth Resources & Applied Geology)	50		

### -: Note :-

प्रत्येक वर्ष के विद्यार्थियों हेतु पाठ्यक्रम में उल्लेखित भूवैज्ञानिक क्षेत्रीय अध्ययन अनिवार्य होगा।

  
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कक्षा / Class- B.Sc-II  
Paper –I  
शैलिकी  
(PETROLOGY)

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- इकाई—01
- (i) मैग्मा; परिभाषा, उत्पत्ति एवं संगठन
  - (ii) बॉवेन की अभिक्रिया श्रेणी, मैग्मीय विभेदन एवं स्वांगीकरण
  - (iii) तंत्र, प्रावस्था एवं घटक, उष्मागतिकी के सिद्धांत, एकघटकीय (सिलिका) द्विघटकीय ऐल्बर्ट-एनॉर्थाइट तथा डायोप्साइड-एनॉर्थाइट एवं त्रिघटकीय सिलिकेट सिस्टम डायोप्साइड-एल्बर्ट-एनॉर्थाइट क्रिस्टलीकरण, प्रावस्था संतुलन
  - (iv) आग्नेय शैलों का गठन, संरचनायें एवं वर्गीकरण
  - (v) आग्नेय शैलों का रूप
- इकाई—02
- (i) दिक्काल में शैल-संलग्नता, शैल-ग्रंथियों की अवधारणा
  - (ii) अम्लीय आग्नेय शैलों का शिला विवरणात्मक अध्ययन
  - (iii) क्षारीय आग्नेय शैलों का शिला-विवरणात्मक अध्ययन
  - (iv) अल्पसिलिक आग्नेय शैलों का शिलाविवरणात्मक अध्ययन
  - (v) अत्यल्पसिलिक आग्नेय शैलों का शिलाविवरणात्मक अध्ययन
- इकाई—03
- (i) अवसाद की उत्पत्ति, परिवहन एवं निक्षेपण
  - (ii) अवसाद निक्षेपण की वायूढ़, जलोढ़, तटीय, एवं गंभीर समुद्री वातावरण की गतिकी
  - (iii) अवसादी संलक्षणाओं की अवधारणा
  - (iv) डायजिनेसिस की अवधारणा
  - (v) अवसादी शैलों का गठन एवं संरचनायें
- इकाई—04
- (i) अवसादी शैलों का वर्गीकरण
  - (ii) अवसादी शैलों की शैलिकी : रूडेशियस, एरेनेशियस, केल्केरियस अवसादी शैल
  - (iii) कायान्तरण: परिभाषा एवं कारक, संलक्षणा, कायान्तरण श्रेणी
  - (iv) कायान्तरित शैलों का गठन, संरचना एवं वर्गीकरण
  - (v) कायान्तरण प्रक्रियाओं की साम्य एवं असाम्य अभिक्रियायें

  
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- इकाई—05 (i) पैराजिनेटिक—आरेख: प्रक्षेपीय विश्लेषण, ए.सी.एफ. एवं ए.के.एफ. आरेख  
(ii) मृण्मय अवसादों का प्रगामी कायान्तरण  
(iii) अशुद्ध चूना पत्थरों का प्रगामी—उष्मागतिक कायान्तरण  
(iv) अल्प सिलिक शैलों का प्रगामी उष्मागतिक कायान्तरण  
(v) भारत का शैलिकीय—प्रादेशिक विभाजन

### **प्रायोगिक कार्य—**

- (1) आग्नेय, अवसादी एवं कायान्तरित शैलों के विभिन्न रूपों एवं संरचनाओं को रेखाचित्र की सहायता से प्रदर्शित करना।
- (2) विभिन्न आग्नेय शैलों का स्थूलदर्शी अध्ययन एवं सूक्ष्मदर्शी अध्ययन
- (3) विभिन्न अवसादी शैलों का स्थूलदर्शी एवं सूक्ष्मदर्शी अध्ययन
- (4) विभिन्न कायान्तरित शैलों का स्थूलदर्शी एवं सूक्ष्मदर्शी अध्ययन
- (5) भारत के शैलिकीय प्रदेशों का मानचित्र में प्रदर्शन
- (6) नार्म कैलकुलेशन

### **Suggested Readings:-**

- |  |   |  |
|--|---|--|
| (1) शैलिकी के सिद्धान्त                                | — | डॉ. अंबिका प्रसाद अग्रवाल              |
| (2) शैलिकी के सिद्धान्त                                | — | ए. जी. झिंगरन                          |
| (3) Principles of petrology                            | - | G.W. Tyrell                            |
| (4) Petrology  | - | H. William, F.J. Turner & E.M. Gilbert |
| (5) Petrology of igneous & metamorphic rocks of India- |   | S.C. Chattarjee                        |
| (6) A text book of sedimentary petrology               | - | Verma & Prasad                         |
| (7) Metamorphism & Metamorphic rocks of India-         |   | S. Ray                                 |
| (8) Sedimentary rocks                                  | - | F.J. Pettijohn                         |
| (9) Introduction of sedimentology                      | - | S. Sengupta                            |
| (10) Sedimentary Environment                           | - | H.G. Readings                          |

  
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- Unit:1**
- (i) Magma, definition, origin & composition
  - (ii) Bowen's reaction series, magmatic differentiation & assimilation
  - (iii) System, phases & component, principles of thermodynamics, Crystallisation and phase equilibrium of unicomponent magma:(Silica), Bi-component magma: Albite-Anorthite and Diopside-Anorthite Tri-component magma: Diopside-Albite-Anorthite
  - (iv) Textures, structures & classification of igneous rocks
  - (v) Forms of igneous rocks
- Unit:2**
- (i) Rock association in Time & Space, concepts of rock kindreds
  - (ii) Petrographic studies of Acid igneous rocks.
  - (iii) Petrographic studies of Alkaline igneous rocks
  - (iv) Petrographic studies of Basic igneous rock
  - (v) Petrographic studies of Ultrabasic igneous rocks.
- Unit:3**
- (i) Origin, transportation & deposition of sediments
  - (ii) Dynamics of sedimentary depositional environment; Aeolian, fluvial, coastal and abyssal environment.
  - (iii) Concept of sedimentary facies
  - (iv) Concept of diagenesis
  - (v) Textures & structures of sedimentary rocks.
- Unit:4**
- (i) Classification of sedimentary rocks.
  - (ii) Petrography of sedimentary rock; rudaceous, arenaceous, calcareous sedimentary rocks.
  - (iii) Metamorphism; definition, agents, facies & grade
  - (iv) Textures, structures & classification of metamorphic rocks.
  - (v) Equilibrium & non-equilibrium reactions in metamorphism.
- Unit:5**
- (i) Paragenetic diagrams; projective analysis A.C.F & A.K.F. diagrams
  - (ii) Progressive metamorphism of Argillaceous rocks.
  - (iii) Progressive dynamo-thermal metamorphism of impure limestone.
  - (iv) Progressive dynamo-thermal metamorphism of basic igneous rocks.
  - (v) Petrographic provinces of India.

  
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**Practical:**

- (1) Diagrammatic representation of various forms & structures of igneous, sedimentary & Metamorphic rocks
- (2) Megascopic studies of various sedimentary, metamorphic & igneous rocks.
- (3) Microscopic studies of various sedimentary, metamorphic & igneous rocks.
- (4) Norm calculation
- (5) Diagrammatic representation of petrographic provinces of India in outline map of India.

**Suggested Readings:-**

- (1) शैलिकी के सिद्धान्त – डॉ. अंबिका प्रसाद अग्रवाल
- (2) शैलिकी के सिद्धान्त – ए. जी. झिंगरन
- (3) Principles of petrology - G.W. Tyrell
- (4) Petrology - H. William, F.J. Turner & E.M. Gilbert
- (5) Petrology of igneous & metamorphic rocks of India- S.C. Chattarjee
- (6) A text book of sedimentary petrology - Verma & Prasad
- (7) Metamorphism & Metamorphic rocks of India- S.Ray
- (8) Sedimentary rocks - F.J. Pettijohn
- (9) Introduction of sedimentology - S.Sengupta
- (10) Sedimentary environment - H.G. Readings

  
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कक्षा / Class- B.Sc-II  
Paper –II  
संरचनात्मक भू-विज्ञान  
(STRUCTURAL GEOLOGY)

- इकाई—01
- (1) संरचनात्मक भूविज्ञान की परिभाषा एवं अध्ययन क्षेत्र।
  - (2) शैल दृष्यांशों का अध्ययन। दृष्यांशों पर नति तथा ढाल के प्रभाव।
  - (3) संस्तरण की पहचान। नति एवं नतिलम्ब की माप।
  - (4) क्लाइनोमीटर एवं ब्रन्टन कम्पास।
  - (5) संस्तरों के शीर्ष तथा तल की पहचान।
  - (6) शैलविरूपण की अवधारणा। प्रतिबल तथा विकृति दीर्घवृत्तज की अवधारणा।
- इकाई—02
- (1) वलन की आकारिकी।
  - (2) वलन की ज्यामितिक एवं जननिक वर्गीकरण।
  - (3) स्थलीय तथा भूवैज्ञानिक मानचित्र में वलन की पहचान।
  - (4) दृश्यांशों पर वलन के प्रभाव।
  - (5) वलन क्रियाविधि की प्राथमिक अवधारणा।
- इकाई—03
- (1) भ्रंश आकारिकी। सर्पण और सेपरेशन।
  - (2) भ्रंश का ज्यामितिक एवं जननिक वर्गीकरण।
  - (3) स्थलक्षेत्र तथा भूवैज्ञानिक मानचित्र में भ्रंश की पहचान।
  - (4) दृश्यांशों पर भ्रंश के प्रभाव।
  - (5) भ्रंशान क्रियाविधि की प्राथमिक अवधारणा।
- इकाई—04
- (1) संधि; आकारिकी, संधि का ज्यामितिक एवं जननिक वर्गीकरण।
  - (2) पत्रण की परिभाषिक शब्दावली, प्रकार, उत्पत्ति एवं विशाल संरचनाओं से संबंध।
  - (3) रेखण की परिभाषिक शब्दावली, प्रकार, उत्पत्ति एवं विशाल संरचनाओं से संबंध।
  - (4) लवण गुम्बद,
  - (5) प्लूटान; विवर्तनिकी एवं अभिस्थापन
- इकाई—05
- (1) विषमविन्यास के प्रकार एवं पहचान।
  - (2) पुरान्तशायी एवं नवान्तशायी, अतिव्यापन तथा अपव्यापन।
  - (3) विवर्तनिकी की अवधारणा।
  - (4) प्रायद्वीपीय, सिंधु गंगा के मैदान तथा प्रायद्वीपेत्तर भारत का विवर्तनिकी विन्यास।
  - (5) त्रिविमीय प्रक्षेपण का संरचनात्मक भूविज्ञान में अनुप्रयोग।

  
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## प्रायोगिक कार्य-

- (1) प्राकृतिक संरचनात्मक प्रादर्शों का अध्ययन।
- (2) विभिन्न संरचनाओं का प्रादर्शों के माध्यम से अध्ययन।
- (3) मानचित्र में दृश्यांश को पूरा करना।
- (4) सरल से जटिल संरचनाओं को प्रदर्शित करने वाले मानचित्रों से भूवैज्ञानिक काट बनाना एवं भूवैज्ञानिक इतिहास की विवेचना करना।
- (5) संरचनात्मक भूविज्ञान में स्टिरियोग्राफिक प्रोजेक्शन का अनुप्रयोग।
- (6) सात दिवसीय भूवैज्ञानिक क्षेत्रीय अध्ययन।

  
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Class- B.Sc - II  
Paper –II  
(STRUCTURAL GEOLOGY)

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- Unit:1**
- (i) Definition and scope of Structural Geology. Study of outcrops. Effects of dip and slope on outcrops.
  - (ii) Identification of bedding. Dip and strike measurement.
  - (iii) Clinometer and Brunton compass.
  - (iv) Recognition of top and bottom of beds.
  - (v) Concept of rock deformation. Concept of stress and strain ellipsoids.
- Unit:2**
- (i) Fold morphology.
  - (ii) Geometric and genetic classification of folds.
  - (iii) Recognition of folds in the field and on geological maps.
  - (iv) Effect of folds on outcrops.
  - (v) Elementary idea of mechanics of folding.
- Unit:3**
- (i) Fault morphology. Slip and separation.
  - (ii) Geometric and genetic classification of faults.
  - (iii) Recognition of faults in the field and on geological maps.
  - (iv) Effect of faults on outcrops.
  - (vi) Elementary idea of mechanics of faulting.
- Unit:4**
- (i) Joint morphology; geometric and genetic classification of joints.
  - (ii) Foliation; terminology, kinds, origin and relation to major structures.
  - (iii) Lineation: terminology, kinds, origin and relation to major structures.
  - (iv) Salt domes.
  - (vii) Plutons; tectonics & emplacement.
- Unit:5**
- (i) Types and recognition of Unconformity.
  - (ii) Outlier and inlier. Overlap & offlap.
  - (iii) Concept of tectonics.
  - (iv) Tectonic framework of Peninsula, Indo-Gangetic Plains and Extra-Peninsular India.
  - (v) Stereographic projection & its use in Structural Geology.

  
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## Practical-

- (1) Study of Natural Structures in hand specimens.
- (2) Study of structures with the help of models.
- (3) Completion of outcrops.
- (4) Preparation of geological section from simple to complex geological maps and its interpretation.
- (5) Application of stereographic projection in structural geology.
- (6) Geological excursion for seven days.

## Books recommended:

- (1) संरचनात्मक भूविज्ञान – डॉ.डी.के. श्रीवास्तव
- (2) भूवैज्ञानिक संरचनाएँ – डॉ. भरत सिंह राठौर
- (3) प्रायोगिक भूविज्ञान (भाग-2) – आर.पी. मांजरेकर
- (4) Structural Geology : M.P. Billings.
- (5) Theory of Structural Geology : Gokhale, N.W.
- (6) Exercises on Geological maps and dip-Strike: Gokhale, N.W.
- (7) Outlines of structural Geology: E.S. Hills.
- (8) Structural Geology : Hobbs, Means and Williams.
- (9) Geological maps : Chiplonkar and Pawar.

  
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**B.A./B.Sc. – Second Year**

**Session : 2019-20**

Name of the Subject :- Anthropology  
Paper :- First  
Name of the Paper :- ARCHAEOLOGICAL ANTHROPOLOGY

Total Marks : 50

Pass Marks : 17

**Syllabus**

- UNIT – I Meaning and scope of Archaeological Anthropology, branches of Archaeology: Classical Archaeology, Historical Archaeology, Prehistoric Archaeology and Protohistoric Archaeology. Anthropology as Archaeology. Differences between the Old world and new world Archaeological Traditions. Absolute and Relative Dating.
- UNIT – II Geological time scale. The Great Ice Age  
Stratigraphy and other evidences of Ice Age: River terraces. Moraines etc. Pluvial and interpluvials  
Stone Age tools: Types and Technology.
- UNIT – III Age of Paleolithic savagery:  
European lower Paleolithic period: Stone tools and cultures  
Indian lower Paleolithic period: Sohan Culture & Madrasian Culture.  
European Middle Paleolithic Period: Tools & culture; Flake tool complex in India  
European Upper Paleolithic period; Tools and Culture, main characteristics of the European Paleolithic Home and Cave art and its significance.
- UNIT – IV Mesolithic complex in North Europe. Mesolithic complex in Western Europe, Mesolithic Culture in India. Chief feature of Neolithic revolution. Neolithic complex in India.
- UNIT – V Metal Age: Copper, Bronze and Iron Age  
Urban revolution: General Features  
Indus valley civilization: Main Features, Town Planning, Economic activities, origin and decay

  
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**B.A. /B.Sc. – Second Year**

**Session: 2019-20**

Name of the Subject :- Anthropology  
Paper :- Second  
Name of the Paper :- TRIBAL CULTURE OF INDIA  
Total Marks : 50

Pass Marks : 17

**Syllabus**

- UNIT-I** Define tribe and scheduled tribe. Geographical distribution of Indian tribes and their racial and linguistic classification. Contribution of Anthropology in the study of Indian tribes.  
Sacred complex, Universalisation and parochialisation, Sanskritisation, westernization, dominant caste.  
Tribes and caste, Difference between S.C. and S.T.  
Particularly Vulnerable Tribale Group (PVTG) of Chhattisgarh (Kamar, Birhor, Hill Korwa. Abujhmaria, Baiga)
- UNIT-II** Primitive economy:-  
Stages of tribal economy: Hunting, food gathering, fishing, shifting and settled agriculture.  
Concept of Property and ownership in tribal societies  
Problems of tribal people: land alienation, bonded labour, indebtedness, shifting cultivation, irrigation, Unemployment, agricultural labour; Forest and Tribals  
New economic anthropology: Exchange- Gifts, barter, trade, ceremonial exchange and market economy
- UNIT-III** The problems of culture contact: Problems due to urbanization and industrialization, Regionalism  
Tribal religion: origin & function, animism, totemism.  
Concept and practices of Magic and witchcraft, shamanism, head hunting.
- UNIT-IV** Political organisation of Indian tribes: Distinction between state and stateless society, law in primitive society  
Social organization of Indian Tribes: Matriarchal and patriarchal family,. Lineage and clan, Ways of acquiring mates in tribal societies.  
Youth dormitories: Type, organisation and functions.
- UNIT-V** Tribal development: History of tribal development, the constitutional safeguards for the scheduled tribes.  
Tribal problem: isolation, migration, acculturation, detribalization.  
Policies, plans and programmes of tribal development and their implementation. Tribal revolts in India.  
Contributions of anthropology to tribal development.  
Response of the tribal people for development programs of government and NGO

  
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Recommended Readings:

1. Chaudhary, Bhudadeb (Ed.). Tribal Development in India.
2. Elwin, V.A. Philosophy for NEFA.
3. Haimendorf. The Tribes of India: Struggle for survival.
4. Shara B.D. Basic Issues in tribal Development.

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## **B.A./B.Sc. – Second Year**

**Session : 2019-20**

Name of the Subject :- Anthropology  
Paper :- Practical  
Name of the Paper :- MATERIAL CULTURE AND RESEARCH TOOLS

Total Marks : 50


Pass Marks : 17

### **OBJECTIVES :**

The objective of this practical course is to introduce the student with the primitive material culture and technology used by primitive man and the student will be introduced with various techniques commonly used by social Anthropology.

### **MATERIAL CULTURE :**

- Part – I. Identification and technological descriptions of the following.
1. Implements for food gathering, hunting, fishing and agriculture
  2. Fire making implements
  3. Types of habitations
  4. Land and water transport
- Part-II Sketching, identification and the description of Paleolithic, Mesolithic and Neolithic tools
- ( It is essential that students should draw at least five tools of each age )
- Part- III Construction of schedule, Geneology and Questionnaire
- Each student should collect information through above tools from 10 Respondents.
- The Student will be required to maintain practical records of all work done in the practical class.

A handwritten signature in blue ink, possibly 'S. Singh', is written above a horizontal line. Below the line, the date '20/06/19' is written in blue ink.

**HEMCHAND YADAV VISHWA VIDYALAYA, DURG (C.G.)**  
**Syllabus for B.A. / B.Sc. Course, 2019-20**  
**Subject: Statistics**

Each year of B.A. /B.Sc. I, II, III shall have two theories and one practical course. All the Theory as well as Practical Examinations will be of 3 hours duration. In each practical examination 10% marks shall be fixed for viva –voce and 20% marks for practical record.

Scheme of Examination

	<b>Title of the paper</b>	<b>MAX. Marks</b>
<b>B.A./B.Sc. I</b>	<b>Paper-I</b> (Code No. 0803): <b>Probability I</b>	50
	<b>Paper-II</b> (Code No. 0804): <b>Descriptive Statistics I</b>	50
	<b>Paper III: Practical-</b> Based on Theory Papers I & II	50
	<b>Total</b>	<b>150</b>
<b>B.A./B.Sc. II</b>	<b>Paper-I</b> (Code No. 0853): <b>Statistical Methods</b>	50
	<b>Paper-II</b> (Code No. 0854): <b>Sampling Theory and Design of Experiments</b>	50
	<b>Paper III: Practical-</b> Based on Theory Papers I & II	50
	<b>Total</b>	<b>150</b>
<b>B.A./B.Sc. III</b>	<b>Paper I</b> (Code No. 0907): <b>Applied Statistics</b>	50
	<b>Paper II</b> ( Code No. 0908): <b>Statistical Quality Control and Computational Techniques</b>	50
	<b>Paper III: Practical-</b> Based on Theory Papers I & II	50
	<b>Total</b>	<b>150</b>

**B.A./B.Sc. –II**  
**Subject: Statistics**  
**Paper-I( Paper Code-0853)**  
**Statistical Methods**

**Unit I**

Sampling from a distribution: Definition of a random sample, simulating random sample from standard distributions (uniform, Normal, Exponential), concept of derived distributions of a functions of random variables, concept of a statistics and its sampling distribution. Point estimate of a parameter. Properties of a good estimator, Concept of bias and standard error of an estimate .Standard errors of sample mean, sample proportion. Sampling distribution of sum of Binomial, Poisson and mean of Normal distributions. Independence of sample mean and variance in random sampling from a Normal distribution (without derivation).

**Unit II**

Statistical tests and interval estimation: Null and alternative hypothesis. Types of errors, level of significance, p values, one and two tailed tests, Procedure for testing of hypothesis. Statement of chi-squares, Student's t and F statistics. Testing for the single mean and variance of a univariate normal distribution, testing the equality of two means and testing for the equality of two variances of two univariate normal distributions. Related confidence intervals. Testing for the significance of sample correlation in sampling from bi-variate normal distribution and for equality of means and equality of variances in sampling from bivariate normal populations.

**Unit III**

Large sample tests: use of central limit theorem for testing and interval estimation of a single mean and a single proportion and difference of two means and two proportions, Fisher's Z transformation and its uses. Pearson's chi-square test for goodness of fit and for homogeneity for standard distributions. Contingency table and test of independence in a contingency table.

**Unit IV**

Nonparametric tests: Definition of order statistics and their distributions, Non-parametric tests, Sign test for univariate and bivariate distributions, Wilcoxon test, Mann-Whitney test, Run test, median test and Spearman's rank correlation test.

**Unit V**

Four short notes, one from each unit will be asked. Students have to answer any two.

**REFERENCES**

1. Frund J.E. (2001) Mathematical Statistics, Prentice Hall of India.
2. Goon A.M., Gupta M.K., Das Gupta.B. (1991): Fundamentals of Statistics, Vol.I, World Press, Culcutta.
3. Gupta and Kapoor: Fundamentals of Mathematical Statistics S.Chand & Sons.
4. Hodges, J.L. and Lehman E.L. (1964): Basic Concepts of Probability and Statistics, Holden Day.
5. Mood A.M, Graybill F.A and Boes D.C. (1974): Introduction to the Theory of Statistics, McGraw Hill.

## **ADDITIONAL REFERENCES**

- 1..Bhat B.R., Shrivnakatramana T and Rao Madhava K.S. (1997): A Beginner's Text, Vol. II, New age International (P) Ltd.
2. Rohatgi, V.K. (1967): An Introduction to Probability Theory and Mathematical Statistics, John Wiley & Sons.
3. Snedecor, G.W. and Cochran W.G. (1967): Statistical Methods. Iowa State University Press.

**Paper-II** (Paper Code-0854)  
**Sampling Theory and Design of Experiments**

**Unit I**

Concepts of population and sample, need for sampling, Census and sample survey, Basic concepts in sampling, organizational aspects of survey sampling, sample selection and sample size.  
Some basic sampling methods – simple random sampling (SRS) with and without replacement.

**Unit II**

Stratified random sampling, Systematic sampling, Allocation problems, ratio and regression methods of estimation under SRS.

Non-sampling errors, acquaintance of working (questionnaires, sampling design, methods followed in field investigation, principal findings, etc) of NSSO and other agencies undertaking sample surveys.

**Unit III**

Analysis of variance for one way and two-way classifications. Need for design of experiments, fundamental principal of design, basic designs- CRD, RBD, LSD and their analysis.

**Unit IV**

Missing plot technique. Analysis of co-variance. Factorial experiments :  $2^2$ ,  $2^3$  factorial experiments, illustrations, main effects and interactions, confounding and illustrations. Yates method of finding treatment totals.

**Unit V**

Four short notes, one from each unit will be asked. Students have to answer any two.

**REFERENCES**

1. Cochran W.G. (1977): Sampling Techniques, John Wiley and Sons.
2. Des Raj (2000): Sample Survey Theory, Narosa Publishing House.
3. Murthy M.N.(1967): Sampling Theory and Methods, Statistical Publishing Society, Calcutta.
4. Singh, D. and Chaudhary, F.S. (1986): Theory and analysis of Sample Survey Designs. New Age International Publisher.
5. Sukhatme P.V., Sukhatme B.V., Sukhatme S. and Ashok C.(1984), : Sample Survey Methods and Its Applications, Indian Society of Agricultural Statistics, New Delhi.
6. Das M.N. and Giri (1986) : Design and analysis of experiments, Springer Verlag.
7. Goon A.M., Gupta M.K., Das Gupta B. (1986): Fundamentals of Statistics, Vol.II, World Press, Calcutta.
8. Joshi, D.D.(1987): Linear Estimation and Design of Experiments, Wiley Eastern.
9. Kempthorne O.(1965) : The Design and Analysis of Experiments, Wiley Eastern.

## **Paper III:**

### **Practical : Practicals Based on Paper I & II**

1. drawing random samples from standard univariate discrete and continuous distributions such as Binomial, Poission, Normal, Cauchy and Exponential.
2. Tests of significance based on Student's t, Chi-square, F. Test of significance of sample correlation coefficient. Use of Z Transformation. Testing of equality of means and equality of variance in sampling from bivariate normal.
3. Large sample tests for means and proportions, tests of goodness of fit and independence of attributes in contingency tables.
4. Nonparametric tests: Sign, Run, Median, Wilcoxon, Mann-Whitney tests.
5. Selection of samples and determination of sample size. Simple random sampling, Statified and systematic sampling. Allocation problem in stratified sampling. Ratio and regression methods of estimation.
6. Analysis of variance for one way and two way classifications. Analysis of CRD, RBD and LSD. Analysis of  $2^2$  and  $2^3$  experiments.

**DEFENCE - STUDIES**  
**PAPER - I**  
**WESTERN MILITARY HISTORY**

**(Paper Code - 0867)**

**Note :** The aim of this paper is to give a historical, political & social back ground of the state engaged in the conflicts under study and the factors influencing the development of different forms of warfare and weapons system.

**Note :** Question will be set from each unit there will be only Internal choice.

- UNIT-I**
1. Sun Tzu - Founder of Military Theory and philosophy.
  2. Clausewitz - War and its relationship with politics.
  3. Machiavelli - Renaissance of Art of war.
  4. Jomini - Concept of mass armies.

- UNIT-II**
1. Churchill.
  2. Mahatma Gandhi.
  3. Kautilya.
  4. A. Hitler.

- UNIT-III**
1. Mao Tse Tung.
  2. Che Guevara.
  3. Economic and Psychological war.
  4. Collective Security.

- UNIT-IV**
1. Indo-China War -1962 Causes of war, political & military lesson.
  2. Indo - Pak War -1965 Causes of war, political & military lesson.
  3. Indo - Pak War - 1971 Causes of war, political & military lesson.
  4. Kargil Conflict 1999.

- UNIT-V**
1. Internal & External threats of National Security.
  2. Insurgency and Counter-Insurgency.
  3. Terrorism - Problem and Solution.
  4. Naxalism - Problem and solution.

**REFERENCE BOOKS:**

1. Howard M. : Theory and Practice of war
2. ---, --- : Clausewitz
3. Mao Tse Tung : Guerilla warfare
4. Palit, D.k. : The lightning War Tadit Yudh
5. Mankekar : War of 1971
6. आर.सी. जोहरी : पाश्चात्य सैन्य विचारक
7. शर्मा च निगम : सैन्य विचारक ।

## **PRACTICAL**

There shall be a practical examination of 3.5 hours duration carrying 50 Marks. The division of marks shall be as follow:

- |                                    |   |          |
|------------------------------------|---|----------|
| (a) Exercise based on Map-reading: | : | 15marks  |
| (b) T.W.E.S.T.                     | : | 15marks  |
| (c) Sessional work                 | : | 10marks  |
| (d) Viva-Voce                      | : | 10markss |

### **PART - A**

Map-reading:

1. Scales - Definition, method of expressing, construction of simple, time, diagonal and comparative.
2. Relief and its representation.
3. Slopes and Gradient.
4. Visibility and inter-visibility by Gradient, proportionate and section method.
5. Re-section and inter-section.
6. Grid system-Map reference, Index to map. Four figure and Six figure.

### **PART - B**

7. Organization and equipment of infantry Platoon and Section.
8. Section Formation.
9. Indication of Target by various methods.
10. Fire control order.
11. Patrols.
12. Battle Procedures (ROFT).
13. Verbal Order.
14. Message-Writing.

### **BOOKS RECOMMENDED:**

1. Manual of Map Reading: Landon Her
2. युद्ध स्थल कला : चौ. नरेन्द्र सिंह
3. एन.सी.सी. परिचय : विष्णु कांत शर्मा ।

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# INDUSTRIAL CHEMISTRY

## PAPER – I

(Paper Code - 0871)

M.M. 34

**UNIT-I** Material Science : Mechanical Properties of materials and change with respect to temperature. **02L**

**Material of constructions used in Industry :**

**Metals and Alloys :** Important metals & alloys; iron, copper, aluminium lead, nikel, titanium and their alloys- Mechanical and chemical properties and their applications. **06L**

**Cement :** Types of cement, composition, manufacturing process, setting of cement. **04L**

**Ceramics :** Introduction, Types, Manufacturing process, Applications. Refractories. **04L**

**UNIT-II** Polymeric Mateials : Industrial polymer and comoposite materials- Their constitution, Chemical and physical properties, Industrial applications. **06L**

**UNIT-III** **Glass :** Types, composition, manufacture, physical and chemical properties, Applications. **04L**

**Corrosion :** Various types of corrosion relevant to chemical Industry-Machanism, Preventive methods. **04L**

**UNIT-IV** Pollution : Air, Oxygen, nitrogen cycle, water, Biosphere, flora and fauna, Energy, soil. **05L**

Pollutants and their statutory limits, pollution evaluation methods. **04L**

**UNIT-V** Air pollution-various pollutants. water pollution-organic/inorganic pollutants, Noise pollution, sewage analysis, pesticide pollution, Radiation pollution, green house effect, future. **10L**

**Books Recommended :**

1. Pollution control in chemical & Allied Industries, S.P. Mahajan.
2. Poolution Control in Industries, A Sories of Books by Jones, H.P.
3. Air Pollution - Vol.1 to 4, Editor, STERN, A.C.; Academic Press.
4. Environmental Engineering, G.N. Pandey, Tata McGraw Hill.
5. Homd Book of Air Pollution, A. Parker, Tata McGraw Hill.
6. Science of Ceromic chemical Processing, Hench, L.L.
7. Science of Ceramics, Stewarts, G.H.
8. Chemistry of Cement.
9. Properties of Glass, Morcy, G.W.
10. Chemistry of Glasses, Paul, A.
11. Corrosion, causes & Prevention, Spellur, F.N.

*Abhinav* 24.7.2017    *Abhishek* 24.7.17    *Abhishek* 24/7/17    *Divyanshu* 24/7/17    *Apurva* 24.7.17    *K*

**PAPER - II**  
**(Paper Code - 0872)**

**M.M. 33**

**UNIT-I** Unit processes in organic chemicals manufacture -

**Nitration** : Introduction - Nitrating agents, Kinetics and mechanism of nitration processes such as nitration of :

- i Paraffinic hydrocarbons
- ii. Benzene to nitrobenzene and m-dinitrobenzene
- iii. Chlorobenzene to o and p nitrochloro benzenes.
- iv. Acetanilide to p-nitroacetanilide
- v. Toluene

Continous vs batch nitration.

**12L**

**UNIT-II Helogenation:** Introduction-Kintics of helogenation reactions reagents for elogenation, Helogenation of aromatics-side chain and nuclear helogenations, commercial manufacture of chlorobenzenes, chloral, monochloracetic acid and chloromethanes, dichloro fluormethane.

**09L**

**UNIT-III Sulphonation** : Introduction-sulphonating agents, chemical and physical factors in sulphonation, Kinetics and mechanism of sulphonation reaction, commercial sulfonation of benzene, naphthalene, alkyl benzene, Batch vs continous sultphonation.

**09L**

**UNIT-IV Effluent Treatment and waste Management** : Principles and equipments for aerobic, anaerobic treatment, adsorption, filtration, sedimentation. **09L**

**UNIT-V** Bag fillters, electrostatic precipitator, mist eliminators, wet scrubbers, absorbers, solid waste management, industrial safety. **09L**

**Books Recommended :**

1. Unit process in Organic synthesis P.M. Groggins, McGraw Hill.
2. Effluent Treatment in process Industries - Inst. of Cham. Engg.
3. Effluent Treatment and waste Disposal - Inst. of Chem. Engg.
4. Effluent Treatment and Disposal - Inst. of Chem. Engg.

  
A series of handwritten signatures and dates, likely from examiners, arranged horizontally. The signatures are written in blue ink and include names like 'Abhinav', 'Nishu', 'Dishu', 'Pratik', and 'K'. The dates are '24.7.2017' and '24/7/17'.

**PAPER - III**  
**(Paper Code - 0873)**

**M.M. 33**

**UNIT-I Oxidation :** Introduction-Types of oxidation reactions, oxidizing agents, kinetics and mechanism of oxidation of organic compounds liquid phase oxidation, vapor phase oxidation, commercial manufacture of benzoic acid, maleic anhydride, phthalic anhydride, acrolein, acetaldehyde, acetic acid. **07L**

**UNIT-II Hydrogenation :** Introduction-Kinetics and thermo-dynamics of hydrogenation reactions, catalysts for hydrogenation reactions, hydrogenation of vegetable oil. manufacture of methanol from carbon monoxide and hydrogen, hydrogenation of acids and esters to alcohols, catalytic reforming. **07L**  
Alkylation: Introduction; Types of alkylation, Alkylating agents, Thermodynamics and mechanism of alkylation reactions, manufacture of - alkyl benzenes (for detergent manufacture), ethyl benzene, phenyl ethyl alcohol, N-alkyl anilines (mono and di- methyl anilines) **03L**

**UNIT-III Esterification :** Introduction; Hydrodynamics and kinetics of esterification reactions, Esterification by organic acids, by addition of unsaturated compounds, esterification of carboxy acid derivatives, commercial manufacture of ethyl acetate, dioctyl phthalate, vinyl acetate, cellulose acetate. **04L**

**Amination : (A) By reduction :** Introduction, Methods of reduction-metal and acid, catalytic, sulfide, electrolytic, metal and alkali sulfites, metal hydrides, sodium metal, concentrated caustic oxidation, reduction, commercial manufacture of aniline, m-nitroaniline, p-amino phenol.

**(B) By aminolysis :** Introduction, aminating agents, factors affecting. **09L**

**Hydrolysis :** Introduction; hydrolysing agents, kinetics, thermodynamics and mechanism of hydrolysis. **02L**

**UNIT-IV Process Instrumentation :** concept of measurement and accuracy Principle, construction and working of following measuring instruments.

Temperature : Glass thermometers, bimetallic thermometer pressure spring thermometer, vapour filled thermometers resistance thermometers. radiation pyrometers.

Pressure : Manometers, barometers, bourdon pressure gauge ; bellow type, diaphragm type pressure gauges, macleod gauges, pirani gauges, etc. **12L**

**UNIT-V Liquid level :** Direct-indirect liquid level measurement, Float type liquid level gauge, ultrasonic level gauges; bubbler system, density measurement, viscosity measurement. **07L**

**Books Recommended :**

1. Unit process in organic synthesis, P.M. Groggins, McGraw Hill.
2. Industrial Instrumentation, Bekmen, D.P., John wrleys.
3. Applied Instrumentation in process Industries, Vol. I, II & III, Andrews, W.G., Gulf Publication.
4. Instrumentation and Control for the process Industries, Borer, S. Elsevier Applied Science Publishers.
5. Chemical Enggineer's Hand book, Perry, J.H. and Green, D. McGraw Hill.

**Time : 4 Hours**

**PRACTICALS**

**M.M. 50**

**Unit Process :** One to two examples of each of the following unit processes. Nitration, sulphonation, friedel-crafts reaction, esterification, hydrolysis, oxidation, Halogenation, chloro-sulphonation, reduction, polymerization, reactions of diazonium salts. **Instrumental methods of analysis :** Use of colourimeter pH meter, potentiometer, conductometer, refractometer, polarimeter

**Materialtesting:** Testing of alloys identification of plastics/rubber estimation of yield point, young's modulus, flaredness; Optical, thermal mechanical and electrical properties. **Process Instrumentation :** Transducers of different types. use of Tranducer for measuring flow control. Determinatiaon of flash point and ignition points of liquids.

**Water analysis :** Solid contents, Hardness, COD and other tests as per industrial specifications.

**Flow measuring devices :** Floats Monographs of representative raw materials such as sulphuric acid, toluene, sodium, carbonate, sodium hyroxide, carbon tetrachloride benzoic acid (5-6 compounds). Limit tests for heavy metals Pb, AS, Hg, Fe and ash content.

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**VOCATIONAL COURSE IN ELECTRONIC  
EQUIPMENT MAINTENANCE  
SCHEME OF EXAMINATION**

		<b>Max. Marks</b>	<b>Min. Pass Marks</b>
Paper - I	Operational Principles of Audio	50	17
Paper - II	Microprocessor Based Instrumentation and Control	50	17
	Practicals	50	17

**1. SUBJECT OBJECTIVE :**

The objective of this syllabus is to familiarize students with the fundamentals of electronics and prepares him/her to keep in track with fast change in this field so that he/she is prepared to takenup advance studies or go for self employment. It is proposed to give the students an idea of basics of all the developments in the field of electronics. Efforts are directed to impart some knowledge of computer hardware and software too, which fall in the realu of electronics so that the students become aware of fast changing scene of information superhigh wey also.

**2. JOB POTENTIALS :**

The students in (by) taking up this course may find adequta job- opportunities in industries or manufacturing firms. They may opt for setting up their own small scale industries of electronics, thus enhancing self employment.

3. **Contents :** As per attached syallbus.

4. Subject scheme.

5. On the job training will be imparted in Summer days.

6. As detailed out in the prospectus.

7. As per the draft given in the syllabus.

8. Permissible combination of subject Physics, Mathemetics & Electonic equipment mathematics.

## **PAPER - I**

**(Paper Code - 0859)**

### **OPERATIONAL PRINCIPLES OF AUDIO AND VIDEO EQUIPMENTS**

**M.M. 50**

**UNIT-I** Revision of All and FH, communication bands, signal sources, Basic Principles of propagation of e.m. wave through atmosphere and ionosphere; ground waves, sky waves, space waves, dead zones etc.

**RECEIVING ANTENNAE:** Antenna Parameters like gain, radiation pattern, effective aperture. Ferrite AE. Type of antennae like wire, loop, dish, Yagi, telescopic, their construction and operating principles.

**SUPERHETERODYNE RECEIVERS:** Principles, advantages, block diagram, RF input and AE coupling arrangements, RF amplifiers, mixer, local oscillator, IF amp. Detector, audio amplifier, loud speaker, power requirements, tuning/aligning of receivers, waveforms and voltages at different check points. Circuit reading of various radio sets, repair and trouble shooting, automobile radios.

**UNIT-II ELEMENTS OF A TELEVISION SYSTEM :** Picture transmission, sound transmission, picture reception, sound reception, synchronisation.

**TYPE VIDEO SIGNAL :** Scanning sequence details, sync details of the 625 line system, channel bandwidth, vestigial sideband transmission, reception of vestigial sideband signals, frequency modulation, FH channel bandwidth, channel bandwidth for colour transmission, allocation of frequency bands for television bandwidth for colour transmission, allocation of frequency bands for television signal transmission, television standards.

Picture tubes- monochrome and colour : Beam deflection, face plate, picture tube characteristics, picture tube circuit controls.

**UNIT-III TELEVISION RECEIVERS :** Types of television receivers, receiver sections, video detector, video section fundamentals, video amplifiers-design principles, video amplifier circuits, automatic gain control and noise cancelling circuits, sync separation circuits, sync-processing and AFC circuits, deflection circuits, sound system, RF tuner, video IF amplifiers, receiver power supplies, television receiver antennae, colour television antennae.

**TELEVISION APPLICATIONS :** Television broadcasting, cable television, closed circuit television, theatre television, picture phone and facsimile, video tape recording (VTr, television via satellite, TV games, HDTV, flatpanel TV teleconferencing.

**UNIT-IV TAPE RECORDERS :** Principles of magnetic recording, characteristics of magnetism, the hysteresis loop, recording head, recorded wave-length, response of head during reply, the effect of gap length, low frequency loss, other losses, equalization, the effect of non-linear characteristic of magnification recording bias, A.C. bias, erasing the tape, block diagram of audio tape recorder.

Oscillator, preamplifier, dolby, amplifier, record (play back) head, erase head, tapes (metal polymer), mechanical transport system, stereo recording, double deck, single deck, microphones (RF, Cable), noise, maintenance of mechanical parts, head cleaners, head alignment, graphic equalisers.

**UNIT-V TELEPHONES :** Modulation, demodulation, modem, subscriber frequency allotment, channel organisation, signalling, switching, manual exchanges, STD, ISD, EFABX, Intercom-press on equipment and EPABX, Value added services like FAX E mail.

**MEASURING INSTRUMENTS :** Multimeters analog/digital, oscilloscopes, signal generators, noise and sound level meters, frequency counters, error sources and precautions during measurement.

**GENERAL NOTE :** Familiarisation with catalogues, standard specification, knowledge about companies referring to service manual.

**PAPER - II**  
**MICROPROCESSOR BASED INSTRUMENTATION AND CONTROL**  
**(Paper Code - 0860)**

**M.M. 50**

**UNIT-I MICROCOMPUTER FUNDAMENTALS** : Introduction, simplified microcomputer architecture, simplified memory organization, instruction set, simplified CPU organisation, microcomputer operation, Personal computer organization and Word Processor. Data sheet descriptions, pin diagram and function, microprocessor architecture, using the data/address register, using the stack pointer.

**UNIT-II THE INTEL 8080/8085 MICROPROCESSOR** : Introduction, the 8085 pin diagram and functions, the 8085 architecture, addressing modes, the 8080/8085 instructions set, the 8080/8085 data transfer instructions, the 8080/8085 arithmetic instructions, the 8080/8085 logical instructions, the 8080/8085 stack, I/O, and machine control instructions.

**UNIT-III PROGRAMMING THE MICROPROCESSOR** : Machine and assembly languages, simplified instruction set, instruction set, arithmetic operations, instruction set-logical operations, instruction set-data transfer operations, instruction set branch operations, instruction set-subroutine call and return operations, instruction set-miscellaneous operations, writing a program, addressing modes, program branching, program looping using subroutines.

Programming the 8080/8085 microprocessor : Introduction, straight-line programs, looping programs, mathematical programs.

**UNIT-IV INTERFACING THE MICROPROCESSOR** : Introduction, interfacing with ROM, interfacing with RAM, input/output interfacing basics, interfacing with practical I/O ports, synchronizing I/O data transfers using interrupts. address decoding.

**UNIT-V Application to illustrate the use of microprocessor in :**

- (i) Traffic control
- (i) Temperature control
- (i) Digital clock
- (iv) Stepper motor control
- (v) Washing machine control

## PRACTICALS

A student is required to do atleast 12 experiments in an academic year, and one month Summer Training. The scheme of practical examination will be as follows :

(i) One experiment of 3 hours duration and one Month Summer Training.

(i) Marks

Experiment : 25 Marks

Sessional : 10 Marks

One Month Summer Training : 15 Marks

**Total 50 Marks**

\* The marks for summer training will be awarded by the thachers teadhing the students on the basis of the certificate issued by the external supervisor of the summer training.

## LIST OF PRACTICALS

1. Development of soldering skill by constructing a fewcircuits and testing.
2. PCB making.
3. Study of modulator.
4. Study of oscillator.
5. Tape recorder-testing, assembly and dis-assembly.
6. Radio receiver-testing.
7. Study of PA system and i.s. testing.
8. Study of EPABK, wiring and connectivity with telephone instruments.
9. Familiarisation with 8085 Based microprocessor trainer kit. Location of 8085, 8279, 8253 keyboard, display fields, EPROM Programmer, expansion s lot, TTY and serial lines.
10. Entering and executing an assembly language program, codes for insertion, deletion, memory move, block fill, setting and examining ragisters and memory, single step execution of a program.
11. Writing of a prgram to add. subtract and multiply two numbers stored in memory (nnnn & nnnn \* 1) and place the result in the subsequent memory, (nnn \* 2).
12. Writing of a program to test R.H. for errors by writing O's & 1's in alternaco location and reading it for chaecking.
13. Making of a board with a 3LED's and four switches to connect to the 8085 kit on the expansion slot (8279).
14. Making of a board with a 8 LED's and four switches to connect to the 8-85 kit on the expansion slot (8255).
  - (a) Program the 8255 to glow/switch of LED's.
  - (b) Program the 8255 to switch on and OFF the LED's every few second according to a given pattern (Hint : The pattern can be 01010101 and 10101010 or 001001100, or any other).

**Reference Books:**

1. Fundamentals of acoustics : Kinsler & Frey
2. System trouble shooting : Luces K, Faulken Berry  
Handbook (John Wiley & Sons)
3. Monochrom & Colour Television : P.R. Gulati
4. Television Engineering : Dhake
5. Microprocessor : Gaonkar
6. Microprocessor : B. Ram
7. Microprocessor : Shaum Saries

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**B.SC.-II**  
**COMPUTER SCIENCE**  
**PAPER - I**  
**COMPUTER HARDWARE**  
**(PAPER CODE - 0855)**

**DURATION 3 HOURS**

**MAX.MARKS 50**

**AIM -** The emphasis is on the design concepts & organisational details of the common PC, learning the complicated electronics of the system of the computer Engineers.

**OBJECT OF THE COURSE -**

1. To introduce the overall organisation of the microcomputers.
2. To introduce the common peripheral devices used in computers.
3. To introduce the hardware components, use of micro processor and function of various chips used in microcomputer.

**N.B. :** Since the computer organisation study is very vast & complicated, so the study is restricted to only the description and understanding part, hence the paper setter is requested to keep this important factor in mind.

**UNIT-I CLASSIFICATION AND ORGANIZATION OF COMPUTERS**

Digital and analog computers and its evolution. Major components of digital computers; Memory addressing capability of CPU; word length and processing speed of computers. Microprocessors single chip microcomputers; large and small computers. User interface Hardware software and firmware. multi programming multi user system. Dumb smart and intelligent terminals computer network and multi processing, LAN parallel processing. Flynn's classification of computers. Computer flow and data flow computers.

**UNIT-II CENTRAL PROCESSING UNIT.**

CPU organization, ALU control unit registers. Instructions for INTEL 8085, Instruction word size, Various addressing mode interrupts and exceptions, some special Control signals and I/O devices. Instruction cycle fetch and execute operation, time Diagram, data flow.

**UNIT-III MEMORY OF COMPUTERS.**

Main memory secondary memory, backup memory, cache memory; real and virtual Memory Semiconductor memory. Memory controller and magnetic memory; RAM; disks, optical disks Magnetic bubble memory; DASD, destructive and non destructive. readout. Program of data Memory and MMU.

**UNIT-IV I/O DEVICES.**

I/O devices of micro controller; processors. I/O devices, printer, plotter, other output devices, I/O port serial data transfer scheme, Micro controller, signal processor, I/O processor arithmetic processor.

**UNIT-V SYSTEM SOFTWARE AND PROGRAMMING TECHNIQUE.**

ML, AL, HLL, stack subroutine debugging of programs macro, micro programming, Program Design, software development, flow & chart multi programming, multiuser, multi tasking Protection, operating system and utility program, application package.

**RECOMMENDED BOOKS :**

1. Computer Fundamentals : Architecture and Organization - By B.Ram (Willey Eastern Ltd.)
2. Computers Today - By Donal H. Sanders
3. Computers Fundamental - By Rajaraman.
4. IBM PC - XT Clones - By Govinda Rajalu

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**B.Sc.-II  
PAPER - II  
SOFTWARE  
(Paper Code - 0856)**

AIM - Introduction to the web-language-HTML & problem solving through the concept of object oriented programming.

**OBJECT OF THE COURSE -**

1. To introduce the internet & web related technology & learn the intricacies of web-page designing using HTML.
2. To introduce the object oriented programming concept using C++ language.
3. To introduce the problem solving methodology using the C++ programming features.

**N.B. : Examiners are requested to prepare unit-wise Questions papers.**

**UNIT-I HTML BASICS & WEB SITE DESIGN PRINCIPLES**

Concept of a Web Site, Web Standards, What is HTML? HTML Versions, Naming Scheme for HTML Documents, HTML document/file, HTML Editor, Explanation of the Structure of the homepage, Elements in HTML Documents, HTML Tags, Basic HTML Tags, Comment tag in HTML, Viewing the Source of a web page, How to download the web page source? XHTML, CSS, Extensible Markup Language (XML), Extensible Style sheet language (XSL), Some tips for designing web pages, HTML Document Structure. HTML Document Structure-Head Section, Illustration of Document Structure, <BASE> Element, <ISINDEX> Element, <LINK> Element, <META>, <TITLE> Element, <SCRIPT> Element, Practical Applications, HTML Document Structure-Body Section:-Body elements and its attributes: Background; BackgroundColor; Text; Link; Active Link (ALINK); Visited Link (VLINK); Left margin; Top margin, Organization of Elements in the BODY of the document: Text Block Elements; TextEmphasis Elements; Special Elements — Hypertext Anchors; Character-Level Elements; Character References, Text Block Elements: HR (Horizontal Line); Hn(Headings); P (Paragraph); Lists; ADDRESS; BLOCKQUOTE; TABLE; DIV (HTML3.2 and up); PRE (Preformatted); FORM, Text Emphasis Elements, Special Elements — Hypertext Anchors, Character-Level Elements: line breaks (BR) and Images (IMG), Lists, ADDRESS Element, BLOCKQUOTE Element, TABLE Element, COMMENTS in HTML, CHARACTER Emphasis Modes, Logical & Physical Styles, Netscape, Microsoft and Advanced Standard Elements List, FONT, BASEFONT and CENTER.

**UNIT-II IMAGE, INTERNAL AND EXTERNAL LINKING BETWEEN WEBPAGES**

Netscape, Microsoft and Advanced Standard Elements List, FONT, BASEFONT and CENTER Insertion of images using the element IMG (Attributes: SRC (Source), WIDTH, HEIGHT, ALT (Alternative), ALIGN), IMG (In-line Images) Element and Attributes; Illustrations of IMG Alignment, Image as Hypertext Anchor, Internal and External Linking between Web Pages Hypertext Anchors, HREF in Anchors, Link to a Particular Place in a Document, NAME attribute in an Anchor, Targeting NAME Anchors, TITLE attribute, Practical IT Application Designing web pages links with each other, Designing Frames in HTML. Practical examples.

**UNIT-III INTRODUCTION TO OOP**

Advantages of OOP, The Object Oriented Approach, Characteristics of object oriented languages- Object, Classes, Inheritance, Reusability, Polymorphism and C++.

Function: Function Declaration, Calling Function, Function Defines, Passing Argument to function, Passing Constant, Passing Value, Reference Argument, returning by reference, Inline Function, Function Overloading, Default Arguments in function.

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## UNIT-IV OBJECT CLASSES AND INHERITANCE

Object and Class, Using the class, class constructor, class destructors, object asfunction argument ,copy constructor ,struct and classes , array as class member,Static Class Data, Static Member Functions, , Friend function, Friend class, operatoroverloading. Type of inheritance, Base class, Derive class. Access Specifier:protected. Function Overriding, member function, String, Template Function.

## UNIT-V POINTERS AND VIRTUAL FUNCTION

pointers: & and \* operator pointer variables, .pointer to pointer, void pointer,pointerand array, pointer and function, pointer and string, memory management, new anddelete, pointer to object, this pointer Virtual Function: Virtual Function, Virtualmember function, accesses with pointer,pure virtual function  
File and Stream: C++ streams, C++ Manipulators, Stream class, string I/O, charI/O, Object I/O, I/O with multiple object, Disk I/O,

## RECOMMENDED BOOKS :

- |                                       |   |  |
|---------------------------------------|---|--|
| 1. Introduction to HTML               | : | KamleshAgarwala, O.P.Vyas, Prateek<br>A. Agrawala (KitabMahal Publication) |
| 2. Let us C++                         | : | Y. Kanetkar B.P.B Publication  |
| 3. Programming in C++                 | : | E. Balaguruswami   |
| 4. Mastering in C++                   | : | VenuGopal  |
| 5. Object Oriented Programming in C++ | : | Lafore R, Galgotia Publications.   |

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# Hemchand Yadav Vishwavidyalaya, Durg (C.G.)

Session 2019-20

June 2019 onwards

Class: B.Sc. Electronics

## Scheme of Examination

Paper Code	Course Opted	Title of Course	Theory	Practical	Grand Total	Minimum Passing Marks
<b>First Year</b>						
ELB-101	Core Course	Network Analysis And Analog Electronics	50		100	33
ELB-102	Core Course	Linear and Digital Integrated Circuits	50			
ELB-103P	Core Course Practical/Tutorial	Networks Analysis and Analog Electronics Lab	25	50	50	17
ELB-104P	Core Course Practical/Tutorial	Linear and Digital Integrated Circuits Lab	25			
<b>Second Year</b>						
ELB-201	Core Course	Communication Electronics	50		100	33
ELB-202	Core Course	Microprocessor and Microcontrollers	50			
ELB-203P	Course Practical/Tutorial	Communication Electronics Lab	25	50	50	17
ELB-204P	Course Practical/Tutorial	Microprocessor & Microcontroller Lab	25			
<b>Third Year</b>						
EL301	Skill Enhancement Course	Industrial Electronics	50		100	33
EL302	Skill Enhancement Course	Mobile Application Programming and Introduction to VHDL	50			
EL303P	Skill Enhancement Course Practical	Industrial Electronics Lab	25	50	50	17
EL304P	Skill Enhancement Course Practical	Mobile Application Programming and Introduction to VHDL Lab	25			

# **B . S c . P a r t I I**

## **ELECTRONICS**

### **Paper I**

#### **ELB 201: COMMUNICATION ELECTRONICS**

**Theory:**

**Max. Marks :50**

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#### **Unit-1**

**Electronic communication:** Introduction to communication – means and modes. Need for modulation. Block diagram of an electronic communication system. Brief idea of frequency allocation for radio communication system in India (TRAI). Electromagnetic communication spectrum, band designations and usage. Channels and base-band signals. Concept of Noise, signal-to-noise (S/N) ratio.

#### **Unit-2**

**Analog Modulation:** Amplitude Modulation, modulation index and frequency spectrum. Generation of AM (Emitter Modulation), Amplitude Demodulation (diode detector), Concept of Single side band generation and detection. Frequency Modulation (FM) and Phase Modulation (PM), modulation index and frequency spectrum, equivalence between FM and PM, Generation of FM using VCO, FM detector (slope detector), Qualitative idea of Super heterodyne receiver

**Analog Pulse Modulation:** Channel capacity, Sampling theorem, Basic Principles-PAM, PWM, PPM, modulation and detection technique for PAM only, Multiplexing.

#### **Unit-3**

**Digital Pulse Modulation:** Need for digital transmission, Pulse Code Modulation, Digital Carrier Modulation Techniques, Sampling, Quantization and Encoding. Concept of Amplitude Shift Keying (ASK), Frequency Shift Keying (FSK), Phase Shift Keying (PSK), and Binary Phase Shift Keying (BPSK).

**Optical Communication:** Introduction of Optical Fiber, Block Diagram of optical communication system.

#### **Unit-4**

**Introduction to Communication and Navigation systems:**

**Satellite Communication**– Introduction, need, Geosynchronous satellite orbits, geostationary satellite advantages of geostationary satellites. Satellite visibility, transponders (C - Band), path loss, ground station, simplified block diagram of earth station. Uplink and downlink.

## Unit-5

**Mobile Telephony System** – Basic concept of mobile communication, frequency bands used in mobile communication, concept of cell sectoring and cell splitting, SIM number, IMEI number, need for data encryption, architecture (block diagram) of mobile communication network, idea of GSM, CDMA, TDMA and FDMA technologies, simplified block diagram of mobile phone handset, 2G, 3G and 4G concepts (qualitative only). GPS navigation system (qualitative idea only)

### Reference Books:

1. Electronic Communications, D. Roddy and J. Coolen, Pearson Education India.
  2. Advanced Electronics Communication Systems- Tomasi, 6<sup>th</sup> edition, Prentice Hall.
  3. Modern Digital and Analog Communication Systems, B.P. Lathi, 4<sup>th</sup> Edition, 2011, Oxford University Press.
  4. Electronic Communication systems, G. Kennedy, 3<sup>rd</sup> Edn., 1999, Tata McGraw Hill.
  5. Principles of Electronic communication systems – Frenzel, 3rd edition, McGraw Hill
  6. Communication Systems, S. Haykin, 2006, Wiley India
  7. Electronic Communication system, Blake, Cengage, 5<sup>th</sup> edition.
  8. Wireless communications, Andrea Goldsmith, 2015, Cambridge University Press
-

**Paper II**  
**ELB 202 :MICROPROCESSOR ANDMICROCONTROLLER**

**Theory:**

**Max. Marks :50**

**Unit-1**

**Microcomputer Organization:** Input/Output Devices. Data storage (idea of RAM andROM). Computer memory. Memory organization & addressing. Memory Interfacing. Memory Map.

**8085 Microprocessor Architecture:** Main features of 8085. Block diagram. Pin-outdiagram of 8085. Data and address buses. Registers. ALU. Stack memory. Program counter.

**Unit-2**

**8085 Programming :**Instruction classification, Instructions set (Data transfer includingstacks. Arithmetic, logical, branch, and control instructions). Subroutines, delay loops. Timing & Control circuitry. Timing states. Instruction cycle, Timing diagram of MOV and MVI. Hardware and software interrupts.

**Unit-3**

**8051 microcontroller:** Introduction and block diagram of 8051 microcontroller,architecture of 8051, overview of 8051 family, 8051 assembly language programming, Program Counter and ROM memory map, Data types and directives, Flag bits and Program Status Word (PSW) register, Jump, loop and call instructions.

**Unit 4**

**8051 I/O port programming:** Introduction of I/O port programming, pin out diagram of8051 microcontroller, I/O port pins description & their functions, I/O port programming in 8051 (using assembly language), I/O programming: Bit manipulation.

**8051 Programming:** 8051 addressing modes and accessing memory locations usingvarious addressing modes, assembly language instructions using each addressing mode, arithmetic and logic instructions,

**Unit 5**

8051 programming in C: for time delay & I/O operations and manipulation, for arithmetic and logic operations, for ASCII and BCD conversions.

**Introduction to embedded system:** Embedded systems and general purpose computersystems. Architecture of embedded system. Classifications, applications and purpose of embedded systems.

## Reference Books:

1. Microprocessor Architecture Programming & applications with 8085, 2002, R.S. Goankar, Prentice Hall.
  2. Embedded Systems: Architecture, Programming & Design, Raj Kamal, 2008, Tata McGraw Hill
  3. The 8051 Microcontroller and Embedded Systems Using Assembly and C, M.A. Mazidi, J.G. Mazidi, and R.D. McKinlay, 2<sup>nd</sup> Ed., 2007, Pearson Education India.
  4. Microprocessor and Microcontrollers, N. Senthil Kumar, 2010, Oxford University Press
  5. 8051 microcontrollers, Satish Shah, 2010, Oxford University Press.
  6. Embedded Systems: Design & applications, S.F. Barrett, 2008, Pearson Education India
  7. Introduction to embedded system, K.V. Shibu, 1<sup>st</sup> edition, 2009, McGraw Hill
  8. Embedded Microcomputer systems: Real time interfacing, J.W. Valvano 2011, Cengage Learning
-

## **ELECTRONICS LABORATORY**

*The scheme of practical examination will be as follows-*

<b>Experiment</b>	--	<b>30</b>
<b>Viva</b>	--	<b>10</b>
<b>Sessional</b>	--	<b>10</b>
<b>Total</b>	--	<b>50</b>

### **ELB 203P: COMMUNICATIONELECTRONICS LAB (Hardware and Circuit Simulation Software) 60 Lectures Max.Marks:25**

1. To design an Amplitude Modulator using Transistor
2. To study envelope detector for demodulation of AM signal
3. To study FM - Generator and Detector circuit
4. To study AM Transmitter and Receiver
5. To study FM Transmitter and Receiver
6. To study Time Division Multiplexing (TDM)
7. To study Pulse Amplitude Modulation (PAM)
8. To study Pulse Width Modulation (PWM)
9. To study Pulse Position Modulation (PPM)
10. To study ASK, PSK and FSK modulators

#### **Reference Books:**

1. Electronic Communication systems, G. Kennedy, 1999, Tata McGraw Hill.
2. Electronic Communication system, Blake, Cengage, 5th edition.

**ELB 204P: MICROPROCESSOR AND MICROCONTROLLER**  
**LAB(Hardware and Circuit Simulation Software)**

**Max.Marks:25**

**At least 06 experiments each from Section-A and Section-B**

***Section-A: Programs using 8085 Microprocessor***

1. Addition and subtraction of numbers using direct addressing mode
2. Addition and subtraction of numbers using indirect addressing mode
3. Multiplication by repeated addition.
4. Division by repeated subtraction.
5. Handling of 16-bit Numbers.
6. Use of CALL and RETURN Instruction.
7. Block data handling.
8. Other programs (e.g. Parity Check, using interrupts, etc.).

***Section-B: Experiments using 8051 microcontroller:***

1. To find that the given numbers is prime or not.
2. To find the factorial of a number.
3. Write a program to make the two numbers equal by increasing the smallest number and decreasing the largest number.
4. Use one of the four ports of 8051 for O/P interfaced to eight LED's. Simulate binary counter (8 bit) on LED's .
5. Program to glow the first four LEDs then next four using TIMER application.
6. Program to rotate the contents of the accumulator first right and then left
7. Program to run a countdown from 9-0 in the seven segment LED display.
8. To interface seven segment LED display with 8051 microcontroller and display 'HELP' in the seven segment LED display.
9. To toggle '1234' as '1324' in the seven segment LED display.
10. Interface stepper motor with 8051 and write a program to move the motor through a given angle in clock wise or counter clockwise direction.
11. Application of embedded systems: Temperature measurement & display on LCD

**Reference Books:**

1. Microprocessor Architecture Programming & applications with 8085, 2002, R.S. Goankar, Prentice Hall.
2. Embedded Systems: Architecture, Programming & Design, Raj Kamal, 2008, Tata McGraw Hill
3. The 8051 Microcontroller and Embedded Systems Using Assembly and C, M.A. Mazidi, J.G. Mazidi, and R.D. McKinlay, 2<sup>nd</sup> Ed., 2007, Pearson Education India.
4. 8051 microcontrollers, Satish Shah, 2010, Oxford University Press.
5. Embedded Microcomputer systems: Real time interfacing, J.W. Valvano 2011, Cengage Learning.

**B.Sc.-II**  
**INFORMATION TECHNOLOGY**  
**PAPER - I**  
**DIGITAL CIRCUITS & COMPUTERH/W**  
**(Paper Code - 0874)**

**UNIT-I(A) Number Systems :**

Octal and hexadecimal number, decimal rep., complements, addition, subtraction, multiplication, division, fixed point rep, floating point rep., other binary code- gray code, excess 3 gray, 2421, etc. error detection code.

**(B) Boolean Algebra :**

Laws, demorgan'stheorem, Simplification boolean expression & logic diagram, positive & negative logic, K-map and simplification of K-map.

**UNIT-II Combinational circuits :**

Half adder, full adder, flip-flop : SR, JK, D,T, sequential circuits : encoder, decoder, multiplexer, shift register, binary counters, BCD adder.

**UNIT-III Multivibratorcircuits :**

Monostable, astable, bistable, smitt trigger, clocked RS, master-slave flip-flop, edge triggered flip-flop, latch.

Intergratedcircuits :

RTL, DITL, TTL, CMOS, MOS.

**UNIT-IV (A) Central Processing Unit :**

Introduction, register organisation, stack organisation, Instruction formats, Addressing modes.

**(B) I/O Organisation :**

I/O interfaces, Data transfer, types and modes, interrupts, DMA, IOP.

**UNIT-V Memory Organisation :**

Memory hierarcy, main memory, Auxiliary memory, Associative memory, cache memory, virtual memory, memory management techniques.

**REFERENCE TAXT BOOK :**

- |  |   |                 |
|--|---|-----------------|
| 1. Integrated Electronics                  | - | Millman&Halkias |
| 2. Principle of Electronics                | - | V.K. Mehta      |
| 3. Digital Electronics                     | - | R.P. Jain       |
| 4. Computer System Architecture            | - | Morris Mano     |
| 5. Digital Electronics & Computer Hardware | - | Morris Mano     |

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*M*  
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*K. Dubey*  
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*Ho*  
19/6/19

*P. Patel*  
19/6/19

**B.Sc.-II**  
**PAPER - II**  
**(Paper Code - 0875)**

**UNIT-I Introduction to OPP :** Advantages of OPP, the Object oriented approach, characteristics of object oriented languages : object, classes, inheritance, reusability, polymorphism and C++.

**UNIT-II Function :** function declaration, calling function, function definition, passing arguments to function, passing constant, passing value, fegerence argument, returning by reference, inline function, function overloading, default arguments in function.

**UNIT-III Object and Classes,** using the Classes Constructor, class destructor, object as function argument, copy constructor, struct and classes, array as class member, static class data, static member functions, friend function, friend class, operator overloading, type of inheritance, base class derive class, access speceifier, protected, member function.

**UNIT-IV Pointers :** & and \* operator pointer variables, pointer to pointer, void pointer, pointer and array, pointer and functions, pointer and string, memory management, new and delete, pointer to object, this pointer, virtual function : virtual function, virtual member function, accesses with pointer, pure virtual function.

**UNIT-V File and stream :** C++ steams, C++ manipulators, Stream class, string I/O, char I/O, object I/O, I/O with multiple objects, disk I/O.

**REFERENCE TEXT BOOKS:**

- |   |                                    |   |                  |
|---|------------------------------------|---|------------------|
| 1 | Programming in C++                 | - | E. Balaguruswami |
| 2 | Mastering in C++                   | - | VenuGopal        |
| 3 | Object Oriented Programming in C++ | - | Robert Lafore    |
| 4 | Let us C++                         | - | Y. Kanetkar      |

**PRACTICAL WORK**

1. The sufficient Practical work should be done for understanding the paper 2.
2. At least five programs on each unit from unit 2 to unit 5 be prepared.
3. All practical works should be prepared in form of print outs and be valuated while practical examination.

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*K. Dubey*  
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*P. Patel*  
19/6/19

## INDUSTRIAL MICROBIOLOGY

Paper	Title	Time	Marks
First	Environmental Microbiology and Biostatistics	3 hrs.	50
Second	Microbial Physiology and Immunobiotechnology	3 hrs.	50
	PRACTICAL Examination (including sessionals)	4 hrs.	50 (40+10)

**Note :** During Two months Summer Vacation, students will visit some Industries. He/She will submit "Summer Job-Training Report" in B.Sc. IIRD Year Viva Voce Exam.

### PAPER - I

#### ENVIRONMENTAL MICROBIOLOGY AND BIOSTATISTICS

(Paper Code - 0876)

M.M.50

**UNIT-1** Our environment : Soil, water and air. Concept of environment in relation to microbes. Environment included physiological adaptations in microorganisms. Nature of microbial population in soil, water and air. Biogeochemical cycling - Carbon, Nitrogen, Sulphur and Phosphorus.

**UNIT-2** Population interactions : Neutralism, Commensalism, Synergism, Mutualism, Antagonistic relationships. Mycorrhizal associations. VAM and its importance.

**UNIT-3** Nitrogen fixation by symbiotic and non-symbiotic microorganisms. Use of microorganisms as biofertilizers. Mass cultivation of Rhizobium and Azotobacter. Use of blue-green algae as biofertilizers.

**UNIT-4** Liquid waste disposal. Nature of domestic and municipal waste and sewage. Sewage treatment. Solid waste disposal. Methods of disposal of Agricultural waste.

**UNIT-5** Basic idea of probability, normal, binomial and poisson distribution. Mean, Mode and Median. Chi-Square test. Exponential and Logarithmic Functions.

### PRACTICALS

1. Isolation of Microorganisms from Air.
2. Isolation of Microorganisms from Water.
3. Isolation of Microorganisms from soil.
4. Determination of MPN of faecal contaminants in water.
5. Measurement & confirmation of E. coli in water sample.
6. Biochemical tests for identification of enteric bacteria.
7. Study of Rhizobium from root nodules.
8. Study of symbiotic and non-symbiotic blue-green algae.
9. Problems based on the determination of Mean, Median and Mode.
10. Problems on Chi-Square Test.
11. Experiments to demonstrate Symbiotic, Antagonistic activities and relations amongst microbes and their interactions with plants.

  
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### **RECOMMENDED BOOKS :**

1. Introduction to Soil Microbiology by Martin Alexander.
2. General Microbiology by Pelczar, Reid & Chan.
3. Biofertilizers in Agriculture by N.S. Subba Rao.
4. Statistics by Mishra & Mishra.
5. General Microbiology, Vol. II, by Power & Dagainawala.

### **PAPER - II**

### **MICROBIAL PHYSIOLOGY AND IMMUNABIOTECHNOLOGY (Paper Code - 0877)**

**M.M. 50**

**UNIT-1** Diffusion, gaseous exchange, Osmosis, Plasmolysis, Biochemical properties of membranes, Passive and Active transport mechanism. Role of ionophores, group translocation across the membranes.

**UNIT-2** Photosynthetic microbes, Oxygenic and non-oxygenic reaction centre. Electron transport, Photophosphorylation, Calvin Cycle. Photorespiration and its significance. Effect of various factors on rate of photosynthesis.

**UNIT-3** Respiration mechanisms - Breakdown of carbohydrates through glycolysis, Krebs's cycle. Fermentation. Pentose Phosphate Pathway. Fermentation of alcohol, Citric acid and acetic acid.

**UNIT-4** Methanogens and Methylophiles. Sulphur utilizing bacteria. Sulphate reduction pathway. Economic importance of Methylophiles and sulphur utilizing bacteria.

**UNIT-5** History and Scope of immunology, Types of immunity. Antigen-Antibody reactions. Immunoglobulins - Structure and functions. Production of Vaccines and Monoclonal antibodies.

### **PRACTICAL**

1. Isolation of photosynthetic bacteria and cyanobacteria from soil.
2. Isolation and characterisation of Methanogens.
3. Study of Hydrogen-production by bacteria.
4. Measurement of nitrate uptake by microorganisms.
5. Study of nitrate and nitrite reduction by microorganisms.
6. Demonstration of evolution during photosynthesis.
7. Demonstration of plasmolysis, osmosis, active and passive transport mechanism.
8. Testing of Blood Groups.
9. Titration of Antigen and Antibody.
10. Precipitation reaction of antigens and antibodies.

### **BOOK RECOMMENDED :**

1. Cell Biology by Pawar.
2. General Microbiology, Vol. II, by Power and Dagainawala.
3. Immunology by Davis.
4. Immunology by G.P. Talwar.

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**BIOCHEMISTRY**  
**PAPER - I**  
**ENZYMOLGY**

**M.M. 50**

**UNIT-I INTRODUCTION**

History, general characteristics, nomenclature, IUB enzyme classification (rationale, over view and specific examples), significance of numbering system. Definitions with examples of holoenzyme, apoenzyme, coenzymes. cofactors, activators, inhibitors, active site (identification of groups excluded), metallo-enzymes, units of enzyme activity, specific enzymes, Isoenzymes, monomeric enzymes, oligomeric enzymes and multienzyme complexes. Enzyme specificity. Historical perspective, nature of non-enzymatic and enzymatic catalysis. Measurement and expression of enzyme activity-enzyme assays. Definition of IU, Katal, enzyme turn over number and specific activity. Role of non-protein organic molecules and inorganic ions coenzyme, prosthetic groups. Role of vitamins as coenzymes precursors (general treatment).

**UNIT-I ENZYME CATALYSIS**

Role of cofactors in enzyme catalysis : NAD/NADP+, FMN/FAD, coenzyme A, biocytin, cobamide, lipoamide, TPP, pyridoxal phosphate, tetrahydrofolate and metal ions with special emphasis on coenzyme functions. Acid-base catalysis, covalent, proximity and orientation effects, strain and distortion theory. Mechanism of action of chymotrypsin, carboxypeptidase, ribonuclease and lysozyme.

**UNIT- I ENZYME PURIFICATION**

Methods for isolation, purification and characterization of enzymes.

**UNIT-IV ENZYME KINETICS**

Factors affecting enzyme activity : enzyme concentration, substrate concentration, pH and temperature. Derivation of Michaelis-Menten equation for uni-substrate reactions.  $K_m$  and its significance. Line weaver-Burk plot and its limitations. Importance of  $K_m$ . Bi-substrate reactions-brief introduction to sequential and ping-pong mechanism with examples.

Kinetics of zero and first order reactions. Significance and evaluation of energy of activation and free energy.

Reversible and irreversible inhibition, competitive, non-competitive and uncompetitive inhibitions. determination of  $K_m$  &  $V_{max}$  in presence and absence of inhibitor. Allosteric enzymes.

**UNIT-V INDUSTRIAL AND CLINICAL APPLICATION OF ENZYME.**

Immobilization of enzyme and their industrial applications. Production of glucose from starch, cellulose and dextran; use of lactase in dairy industry; production of glucose-fructose syrup from sucrose; use proteases in food, detergent and leather industry; medical application of enzymes. use of glucose oxidase in enzyme electrodes.

*Arjun* 24.7.2017    *Arjun* 24.7.17    *BKS* 24/7/17    *Divyanshu* 24/7/17    *Pranav* 24.7.17    *K*

## PAPER - II

### INTERMEDIARY METABOLISM

M.M. 50

#### UNIT-I INTRODUCTION TO METABOLISM

General features of metabolism, experimental approaches to study metabolism; use of intact organism, bacterial mutants, tissue slices, stable and radioactive isotopes.

#### CARBOHYDRATE METABOLISM

Reactions and energetics of glycolysis. Alcoholic and lactic acid fermentations. Entry of fructose, galactose, mannose etc. Reactions and energetics of TCA cycle. Gluconeogenesis, glycogenesis and glycogenolysis, Reactions and physiological significance of pentose phosphate pathway. Regulation of glycolysis and TCA cycle. Photosynthesis, a brief review.

#### UNIT-II ELECTRON TRANSPORT CHAIN AND OXIDATIVE PHOSPHORYLATION

Structure of mitochondria, sequence of electron carriers, sites of ATP production, inhibitors of electron transport chain. Hypothesis of mitochondrial oxidative phosphorylation (basic concepts). Inhibitors and uncouplers of oxidative phosphorylation. Transport of reducing potentials into mitochondria.

#### UNIT-III LIPID METABOLISM

Introduction, hydrolysis of triacylglycerols, transport of fatty acids into mitochondria.

$\beta$ - oxidation of saturated fatty acids, ATP yield from fatty acid oxidation. Biosynthesis of saturated and unsaturated fatty acids. Metabolism of ketone bodies, oxidation of unsaturated and odd chain fatty acids. Biosynthesis of triglycerides and important phospholipids, glycolipids, sphingolipids and cholesterol. Regulation of cholesterol metabolism.

#### UNIT-IV AMINO ACID METABOLISM

General reactions of amino acid metabolism : transamination, oxidative deamination and decarboxylation. Urea cycle. Degradation and biosynthesis of amino acids. Glycogenic and ketogenic amino acids.

#### UNIT-V NUCLEOTIDE METABOLISM

Sources of the atoms in the purine and pyrimidine molecules. Biosynthesis and degradation of purines and pyrimidines. Regulation of purine and pyrimidine biosynthesis.

#### PORPHYRIN METABOLISM

Biosynthesis and degradation of porphyrins. Production of bile pigments.

  
A series of six handwritten signatures and dates, each on a separate line. From left to right: 1. Signature, 24.7.2017; 2. Signature, 24.7.17; 3. Signature, 24/7/17; 4. Signature, 24/7/17; 5. Signature, 24.7.17; 6. Signature.

## PRACTICAL

1. Separation of Blood Plasma and Serum
  - a. Estimation of proteins from serum by biuret and lowry methods.
  - b. Determination of albumin and A/G ratio in serum.
2. Estimation of bilirubin (conjugated and unconjugated) in serum.
3.
  - i. Estimation of total lipids in serum by vanillin method.
  - ii. Estimation of cholesterol in serum.
4. Estimation of lipoproteins in plasma.
5. Estimation of lactic acid in blood before and after exercise.
6. Estimation of blood urea nitrogen from plasma.
7. Separation and identification of amino acids by (a) paper chromatography and (b) thin-layer chromatography.
8. Separation of polar and non-polar lipids by thin-layer chromatography.
9. Estimation of SGPT and SGOT in serum.
10.
  - a. Assay of serum alkaline phosphatase activity.
  - b. Inhibition of alkaline phosphatase activity by EDTA.
  - c. Effect of substrate concentration on alkaline phosphatase activity and determination of its  $K_m$  value.
11.
  - a. Effect of temperature on enzyme activity and determination of activation energy.
  - b. Effect of pH on enzyme activity and determination of optimum pH.
  - c. Effect of enzyme concentration on enzyme activity.
12.
  - a. Preparation of starch from potato and its hydrolysis by salivary amylase.
  - b. Determination of achromatic point in salivary amylase.
  - c. Effect of sodium chloride on amylases.

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Arshin  
24.7.2017

Arshin  
24.7.17

BUS  
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Divyanshu  
24/7/17

Spandan  
24.7.17

Arshin

# **Syllabus of Biotechnology**

**(B. Sc. II Year)**

**Session**

**2019-2020**

**2020-2021**

  
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**HEMCHAND YADAV VISHWAVIDYALAYA, DURG (C.G.)**

**B.Sc. II  
BIOTECHNOLOGY**

**PAPER – I**

**MOLECULAR BIOLOGY & BIOPHYSICS**

**M.M. 50**

**UNIT-I**

1. Nucleic Acid: Bases, Nucleosides and Nucleotides, DNA and RNA structure.
2. Plasmids.
3. Transposons: Repetitive elements, LINEs & SINEs, Structure of Gene.

**UNIT-II**

1. DNA Replication: Enzymes involved and mechanism of DNA Replication in Prokaryotes.
2. Mutation: Molecular level of Mutation, Types of Mutagens, Spontaneous and Induced Mutation.
3. DNA Repair: NER, BER and Mismatch Repair.

**UNIT-III**

1. Genetic Code: Features, Condon Assignment and Wobble hypothesis.
2. Transcription: Initiation, Elongation and Termination in Prokaryotes.
3. Translation: Initiation, Elongation and Termination Translation machinery in Prokaryotes.  
Operon-Concept of Operator, Regulator, Promoter gene, Inducer and Co-repressor.

**UNIT –IV**

1. Biophysics : Introduction, Scope and Application
2. Principle, Structure, Functions of the following:
  - a. Microscopy
  - b. Colorimeter and Spectroscopy
  - c. Electrophoresis
  - d. Centrifugation
  - e. Chromatography.

**UNIT –V**

1. Radioisotopes techniques: Measurement of radioactivity, Ionization Chambers, Geiger Muller and Scintillation Counter.
2. Autoradiography and DNA Fingerprinting.
3. Biosensor.

  
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## List of Books

1. Gerald Karp - Cell and Molecular biology, 4th Edition (2005).
2. Lewis J.Klein Smith and Valerie M.Kish-Principles of cell and molecular biology-Third Edition (2002)
3. Edition (2002)
4. P.K. Gupta- Cell and molecular biology, Second Edition (2003), Rastogi publications.
5. Richard M-Twyaman-Advanced Molecular Biology, First South Asian Edition (1998), VivaBooks Pvt. Ltd.
6. K. Wilson and J.Walker (2012) Principle and Techniques of Biotechnology and MolecularBiotechnology.
7. Upadhya and Upadhya : Biophysical Chemistry.
8. David, I. Nelson and Michael M.Cox :Lehninger : Principal of Biochemistry 4th Edition. W.H. Freeman and Company, New York.
9. Buchanan, Gruissemen& Jones (2015) Biochemistry & Molecular Biology of Plant, 2<sup>nd</sup> edition.

  
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**B.Sc. II  
BIOTECHNOLOGY**

**PAPER II**

**RECOMBINANT DNA TECHNOLOGY AND GENOMICS**

**M.M. 50**

**UNIT-I**

1. Recombinant DNA technology: General concept. Steps in gene cloning and application.
2. Host controlled Restriction Modification System, Ligases and Polymerases, Klenow fragment, Taq, Pfu polymerase and Nuclease (Endo, Exo and restriction endonuclease).
3. Modification Enzyme (Kinase, Phosphates and terminal deoxynucleotidyl transferase). Reverse Transcriptase.

**UNIT –II**

1. Vectors: Plasmid, Bacteriophages, Cosmid, SV40 and Expression vectors.
2. Gene Library: Genomic and cDNA library.
3. Selection and Screening of Recombinants: Genetic and Hybridization methods.

**UNIT –III**

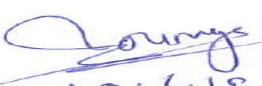
1. PCR: Types of PCR, Steps (Denaturation, Annealing and Extension); Applications, Advantages and Limitation of PCR.
2. Molecular Marker-RFLP, RAPD and Micro array.
3. Human Genome Project.

**UNIT-IV**

1. Basic concept of Gene Transfer Methods: Microinjection, Electroporation, Lipofection and Microprojectile.
2. Gene Therapy: *In vivo* and *Ex vivo*, Germ line and Somatic gene therapy.
3. Basic idea of Stem cell technology: Types of stems cell cultures and their Significance.

**UNIT-V**

1. Introduction to Bioinformatics: History, Objective and Application.
2. Major Bioinformatics Resource – NCBI , Types of Databases (Primary and Secondary Databases) , BLAST and FASTA
3. Basic concept of Genomics and Proteomics

  
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## List of Books

1. B.D. Singh (2004) Biotechnology, Expanding Horizons. First Edition. Kalyani Publishers, Ludhiana.
2. P.K. Gupta (2005) Biotechnology and Genomics, Rastogi Publication, Meerut.
3. Stan bury and Whittaker - Principles of Sterilization techniques, First Indian reprint Edition (1997). Aditya Book (P) Ltd. New Delhi.
4. L.E. Casida (1994)Industrial Microbiology Edition .
5. A.H. Patel (2003) Industrial Microbiology 4th Edition.
6. K.S. Bilgrami and A.K. Pandey(1998) Introduction to Biotechnology Edition 2nd (1998)
7. U Satyanarayan (2005) Biotechnology, First Edition Books and Allied (P) Ltd. Kolkata.
8. Atul kumar and VandanaA.Kumar (2004) Plant Biotechnology and tissue culture, Principle and Perspectives, International Books Distributing Co. Lucknow.
10. S Choudhuri, and DB Carlson (2008) Genomics: Fundamentals and applications, 1st edition.
11. TK Attwood and DJ Parry (2009) Introduction of Bioinformatics.
12. Philip E Bourne Helge Whisking (2003) Structural Bioinformatics.
13. Des Higgins and Willie Taylor (2000) Bioinformatics Sequence, Structure and Databanks.

  
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## List of Practical's

### MOLECULAR BIOLOGY, BIOPHYSICS, RECOMBINANT DNA TECHNOLOGY AND GENOMICS

1. Isolation of DNA from Plant cell.
2. Estimation of DNA by DPA method.
3. Isolation RNA from yeast cells

Experiment based on-

4. Centrifugation
5. Spectrophotometer/Colorimeter
6. Electrophoresis
7. Paper chromatography/TLC

Experiment based on Bioinformatics -

8. Retrieve DNA /Protein sequence from Biological Data Bases (NCBI).
9. Use of tools studied

  
10.6.19

  
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## SCHEME FOR PRACTICAL EXAMINATION

**Time: 4 hrs. M.M.: 50**

- |                                       |          |
|---------------------------------------|----------|
| 1. Experiment based on DNA/RNA        | 10 marks |
| 2. Experiment based on Instruments    | 10 marks |
| 3. Experiment based on Bioinformatics | 10 marks |
| 4. Spotting                           | 10 marks |
| 5. <i>Viva - Voce</i>                 | 05 marks |
| 6. Record / Sessional                 | 05 marks |

  
10.6.19

  
10.6.19

  
10/6/19

  
10.6.19